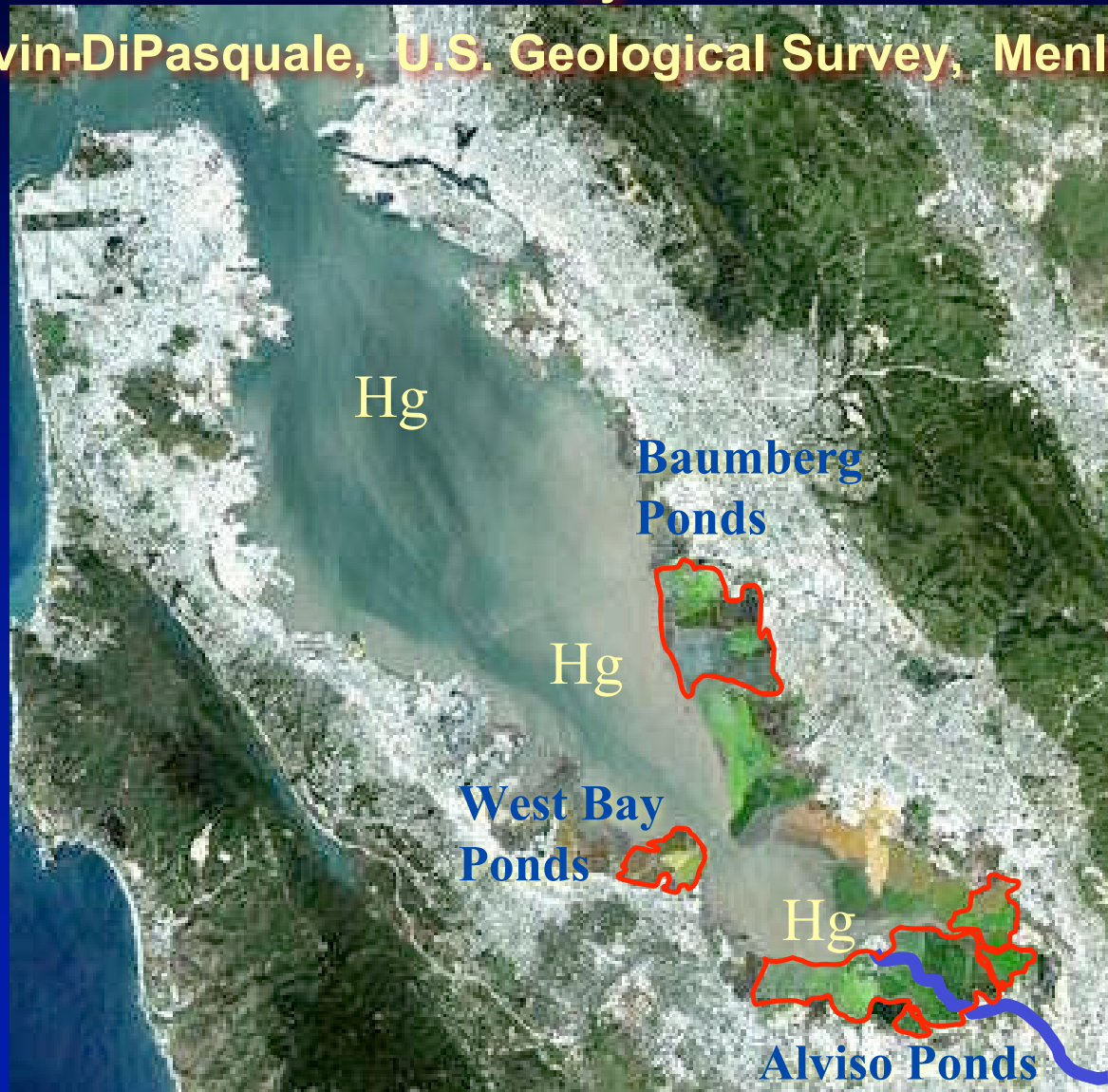


Mercury and the South Bay Salt Pond Restoration Project

By

M. Marvin-DiPasquale, U.S. Geological Survey, Menlo Park, CA



Presentation

➤ What Do We Know?

- ◆ Hg species concentrations in water, sediments and some biota
- ◆ Wetlands are active zones of MeHg production

➤ What Don't We Know?

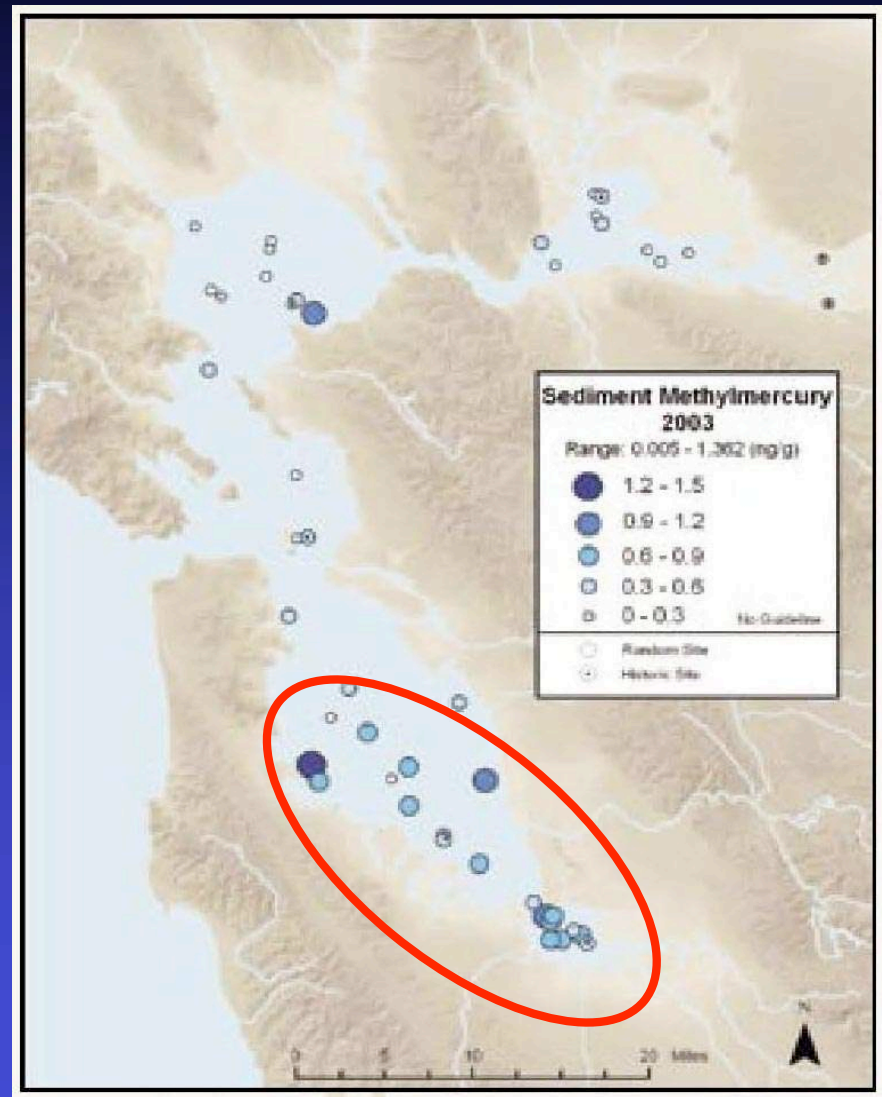
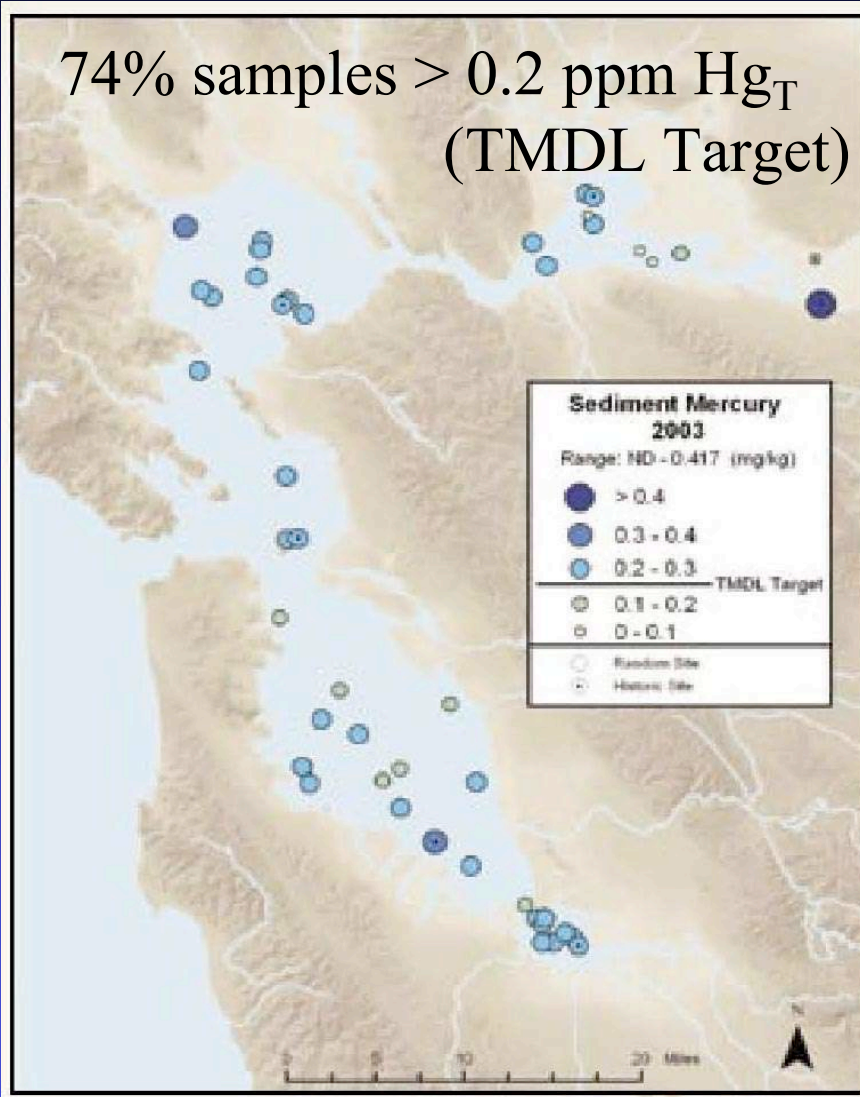
- ◆ Extent of the “Hg Problem” at the Ecosystem Level
- ◆ Habitat specific Processes Controlling:
Hg-methylation, Hg- transport or bioaccumulation

➤ What Are The Implications for SBSP Restoration?

- ◆ Potential for enhanced ‘reactive’ Hg(II) mobilization → MeHg production → bioaccumulation
- ◆ Limited ‘Window of Opportunity’ to mitigate Hg problems → design considerations

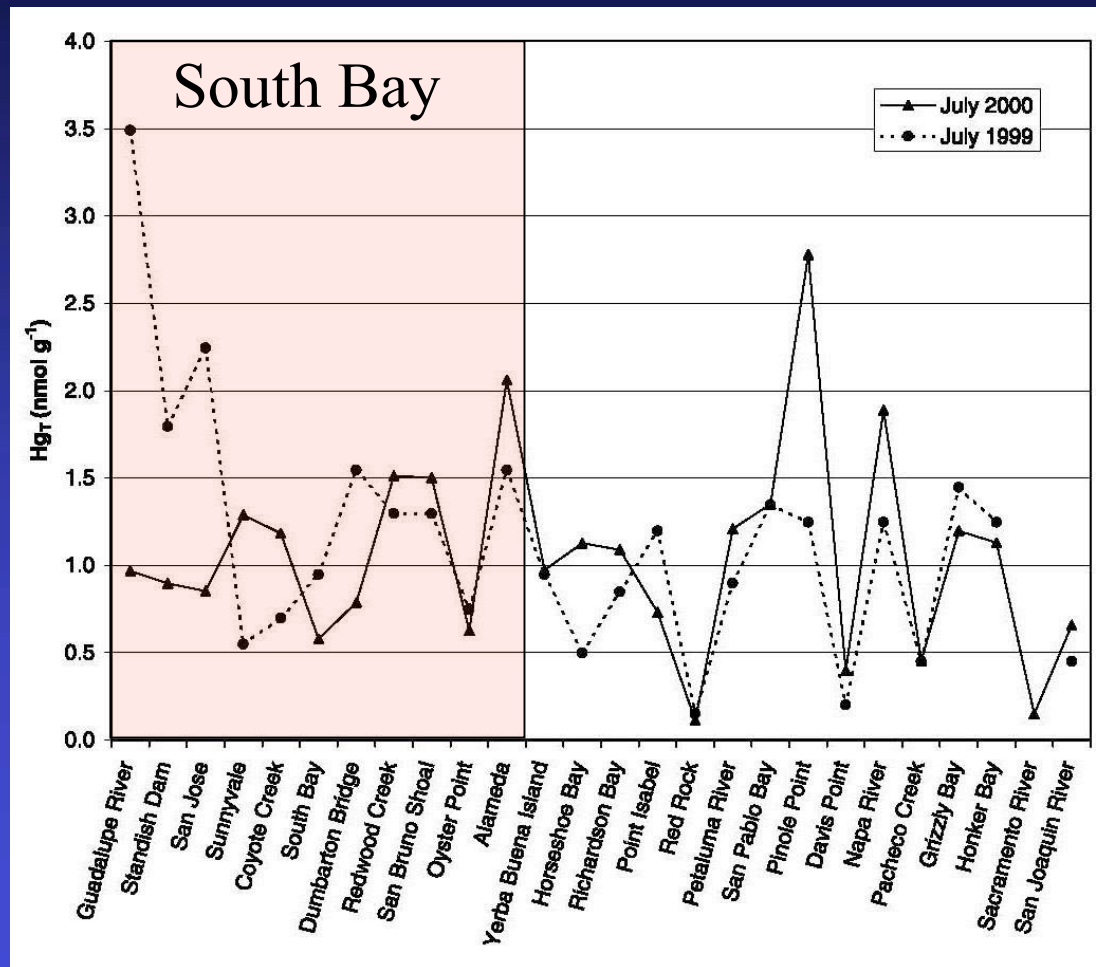
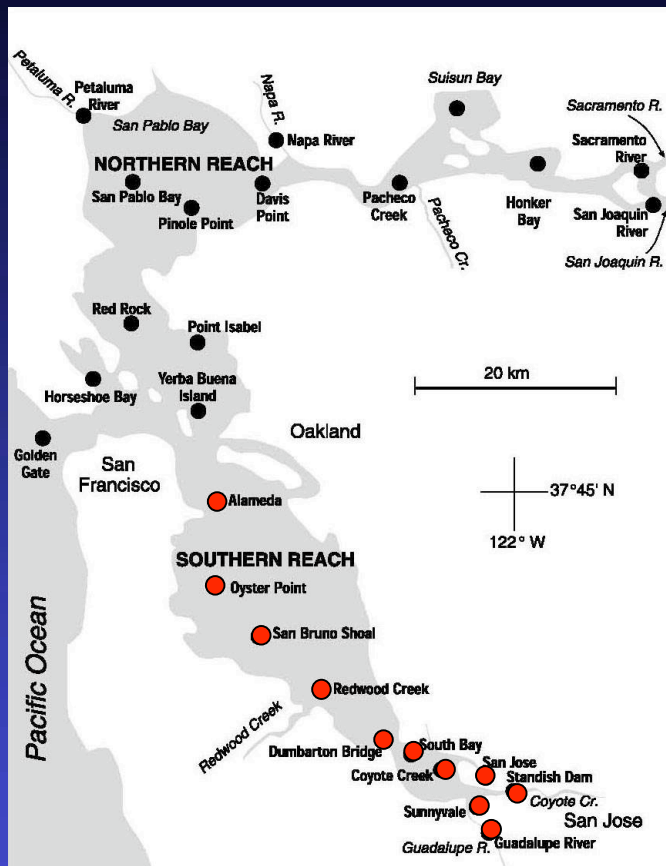
Mercury in Sediment - SFB

74% samples > 0.2 ppm Hg_T
(TMDL Target)



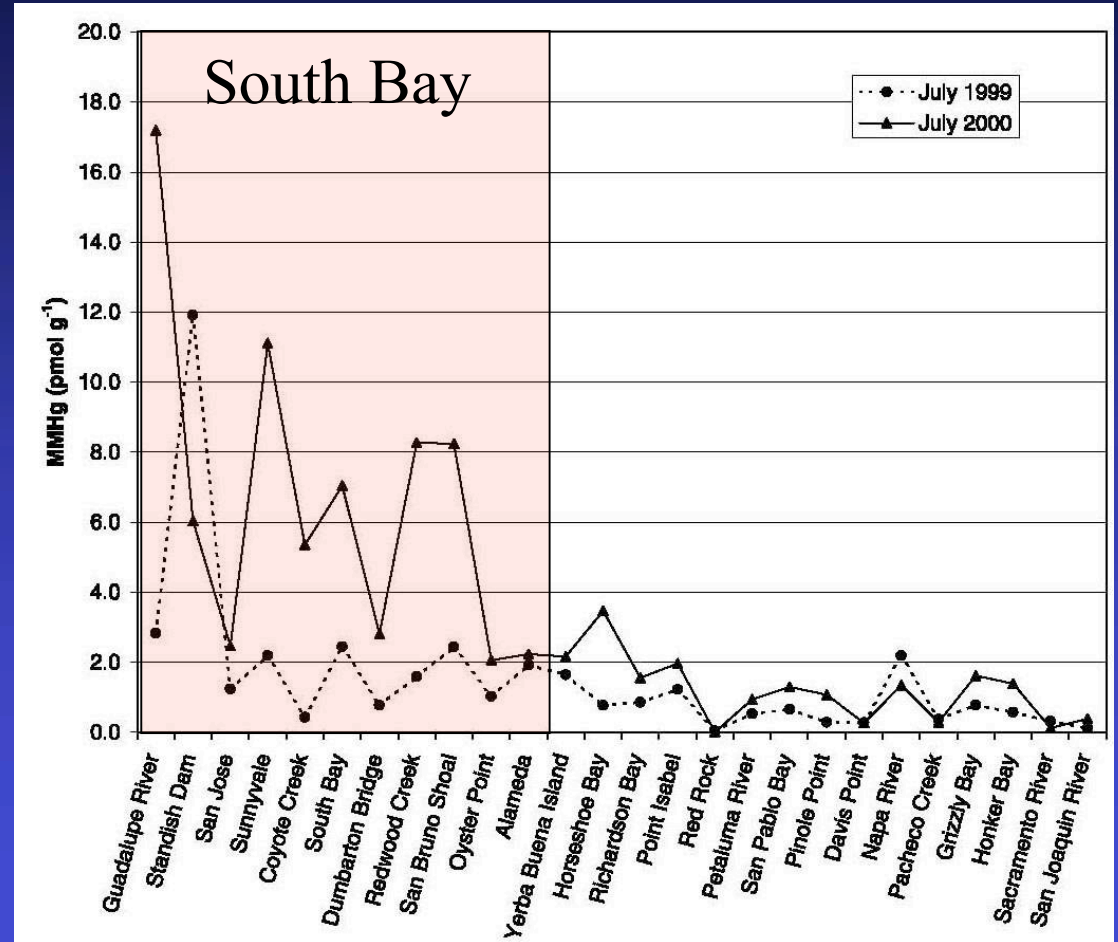
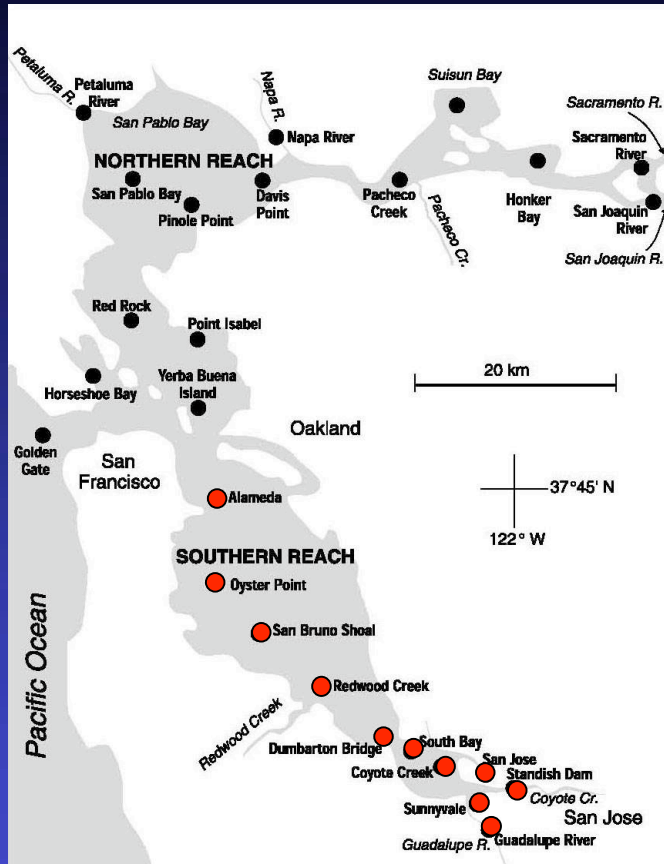
Source: Pulse of the Estuary 2005
Data: Regional Monitoring Program

Total Mercury in Sediment - SFB



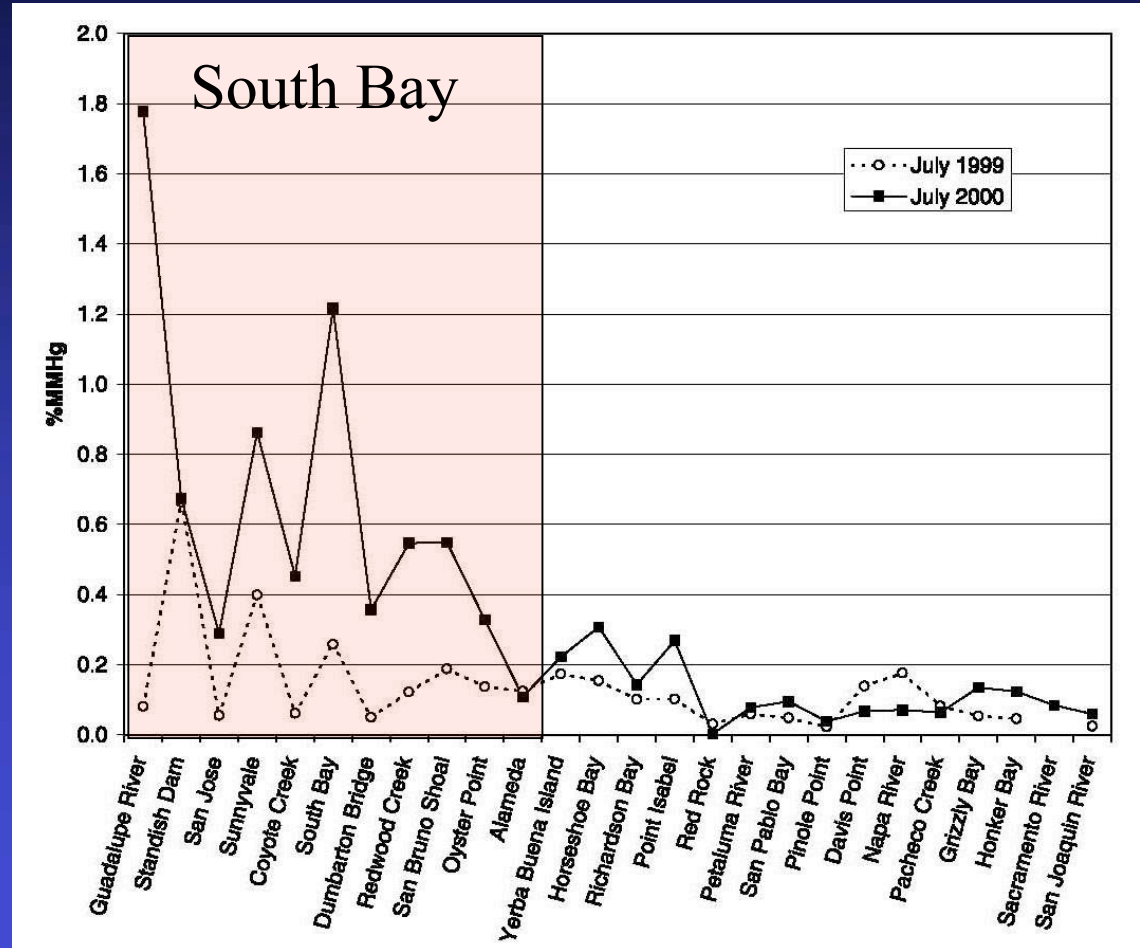
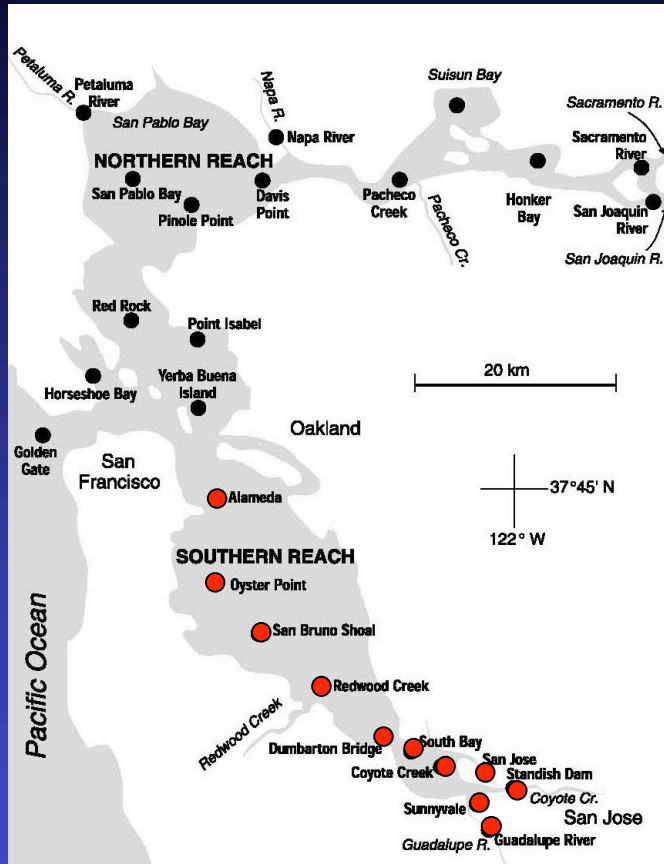
Source: Conaway et al. 2003, Marine Chemistry, v. 80 p. 199–225

Methylmercury in Sediment - SFB



Source: Conaway et al. 2003, Marine Chemistry, v. 80 p. 199–225

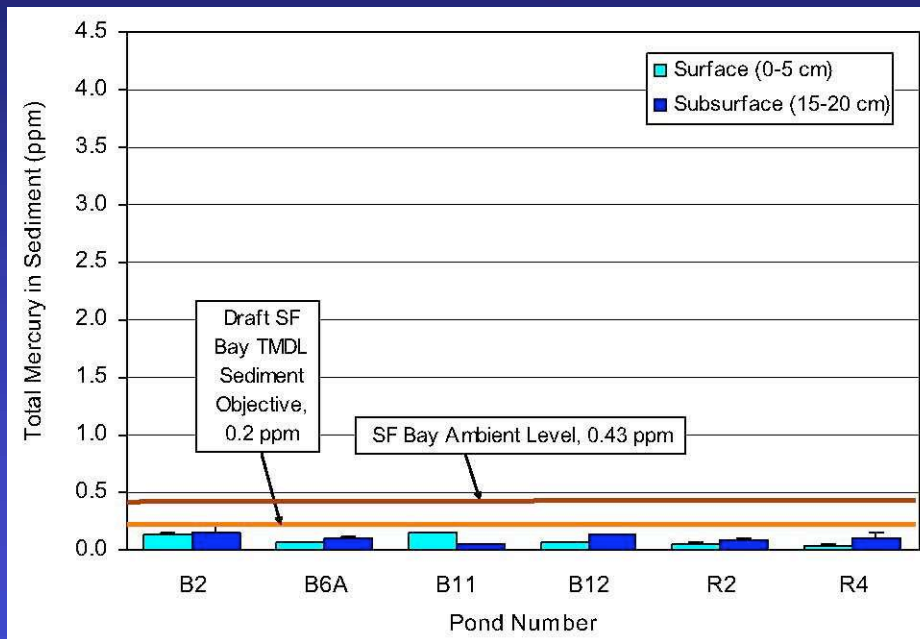
% Methylmercury in Sediment - SFB



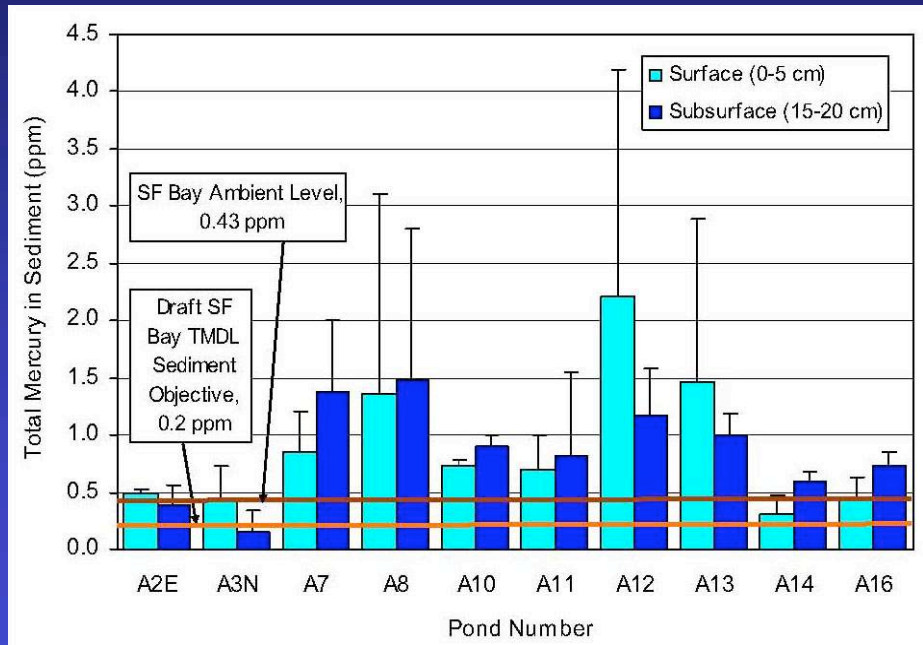
Source: Conaway et al. 2003, Marine Chemistry, v. 80 p. 199–225

Total Mercury in Salt Ponds Sediment

Baumberg/ Eden Landing (B) and Redwood/West Bay (R) Ponds



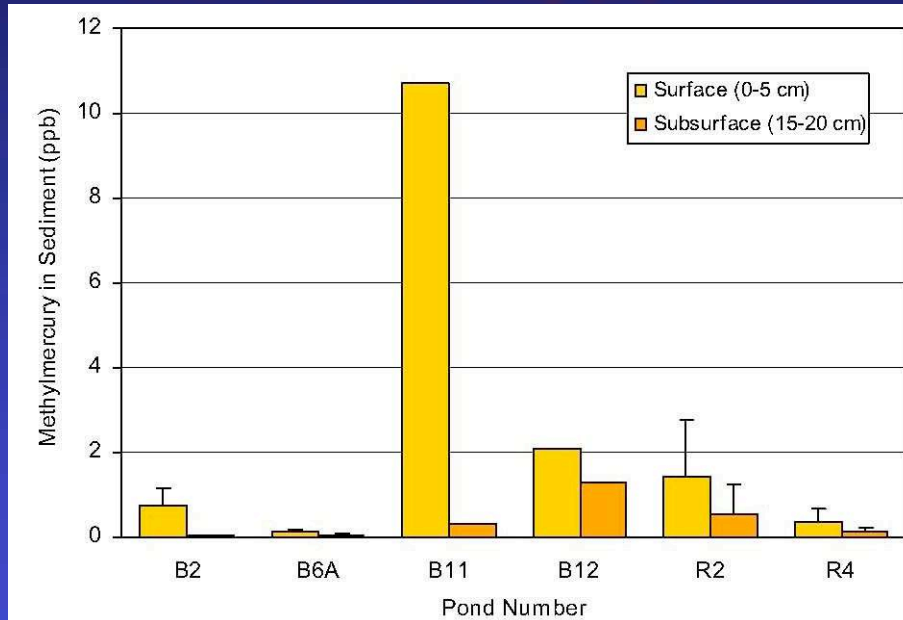
Alviso Ponds



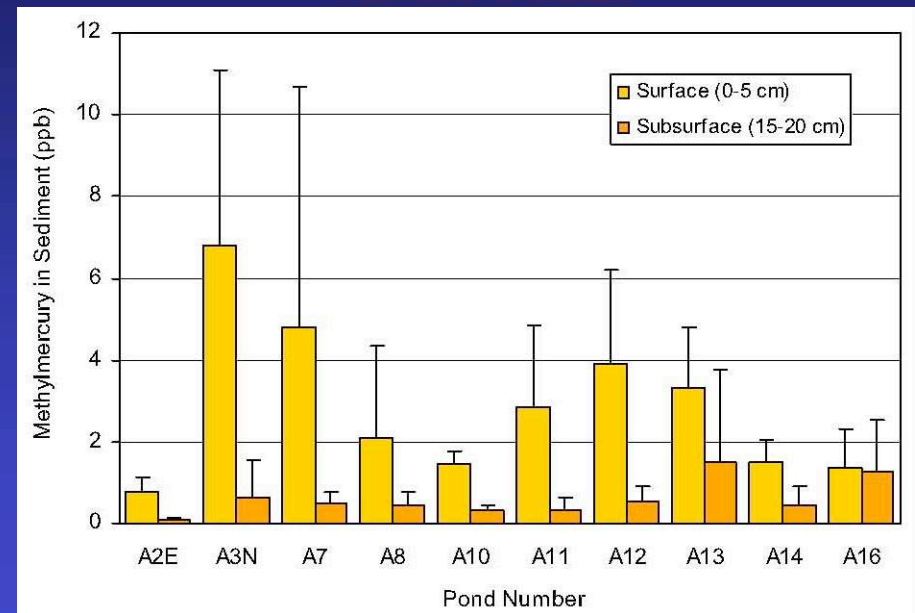
Source: Beutel et al., Mercury Technical Memo, 2004
Data: Keith Miles et al., USGS - WERC

Methylmercury in Salt Ponds Sediment

Baumberg/ Eden Landing (B) and Redwood/West Bay (R) Ponds

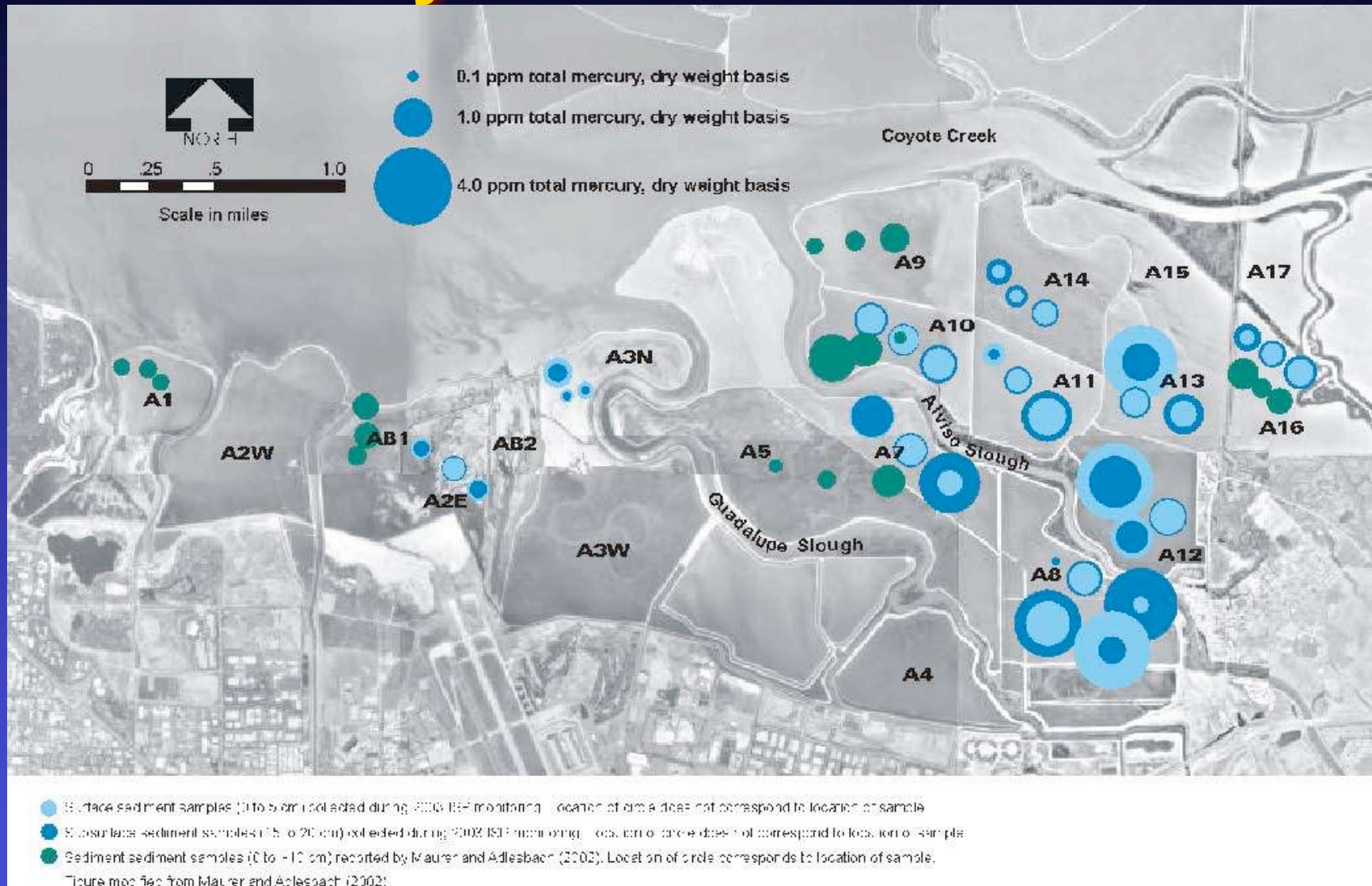


Alviso Ponds



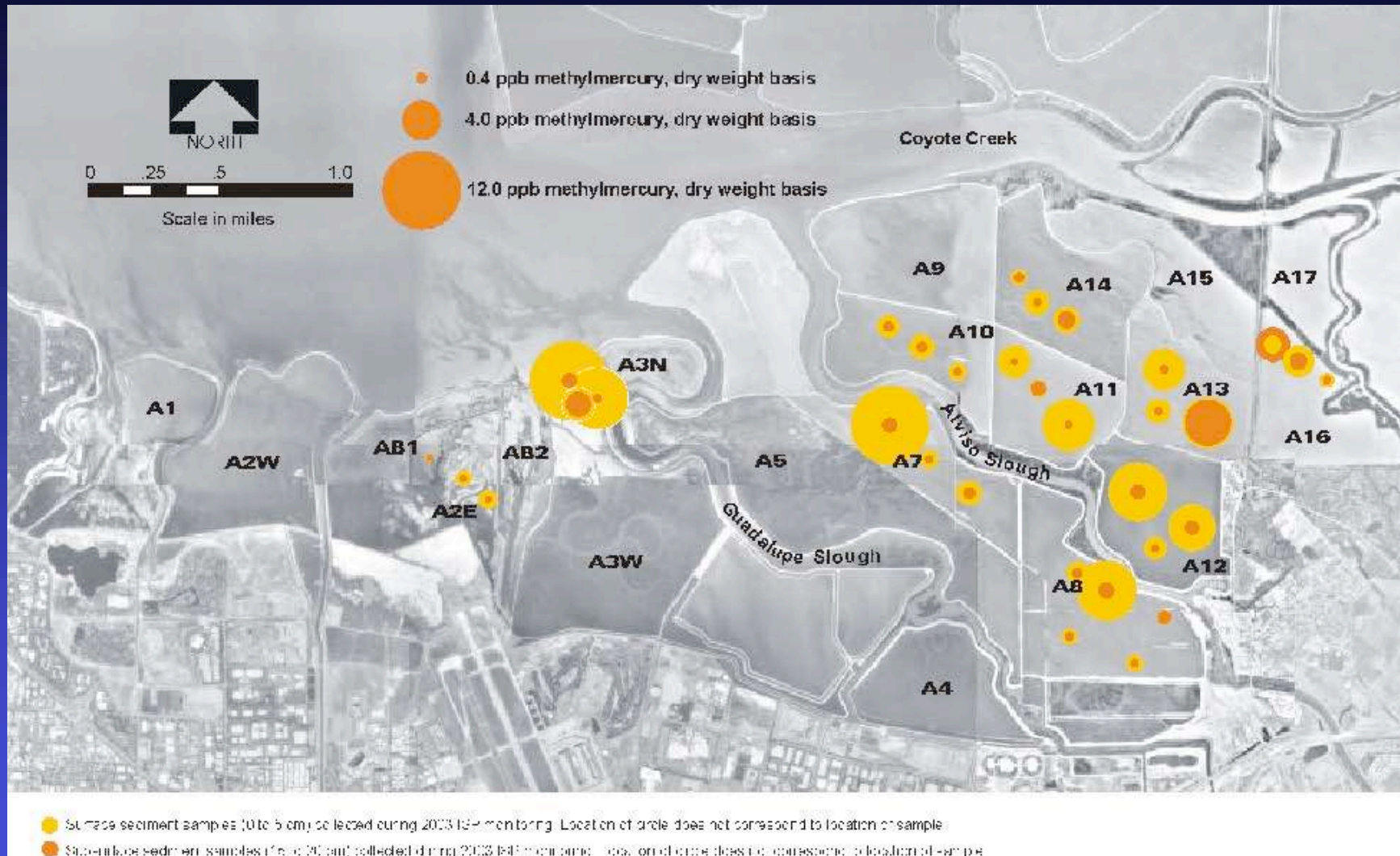
Source: Beutel et al., Mercury Technical Memo, 2004
Data: Miles et al., USGS - WERC

Total Mercury in Alviso Ponds Sediment



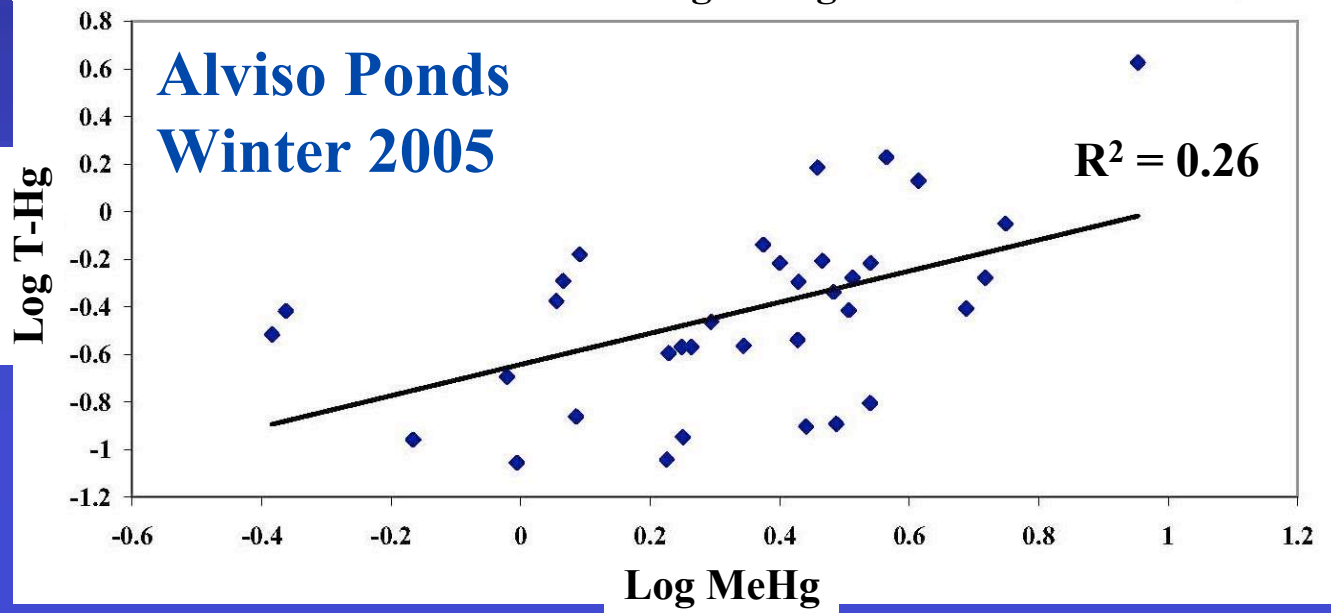
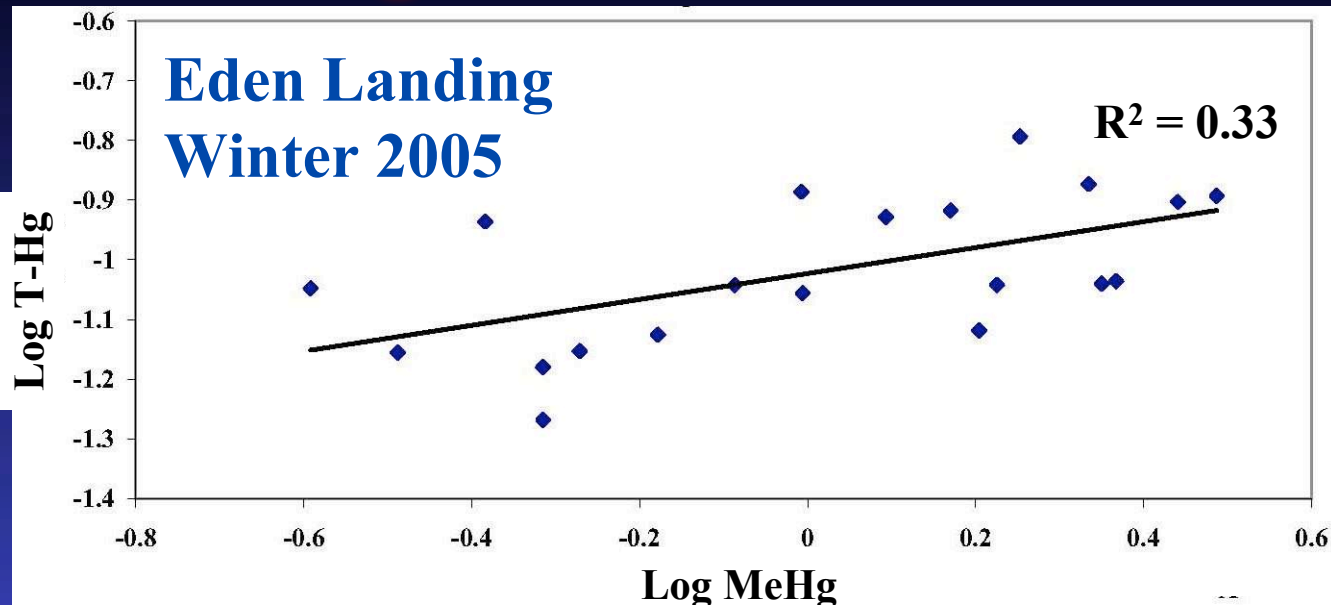
Source: Beutel et al., Mercury Technical Memo, 2004

Methylmercury in Alviso Ponds



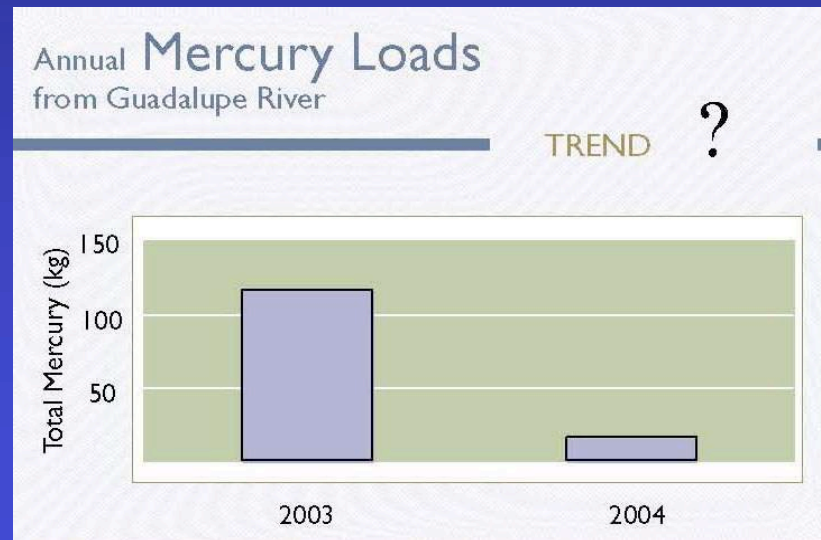
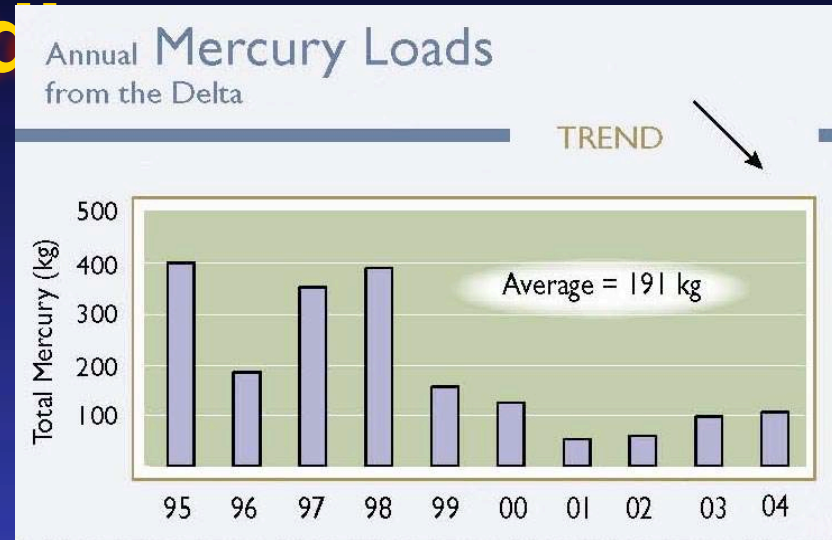
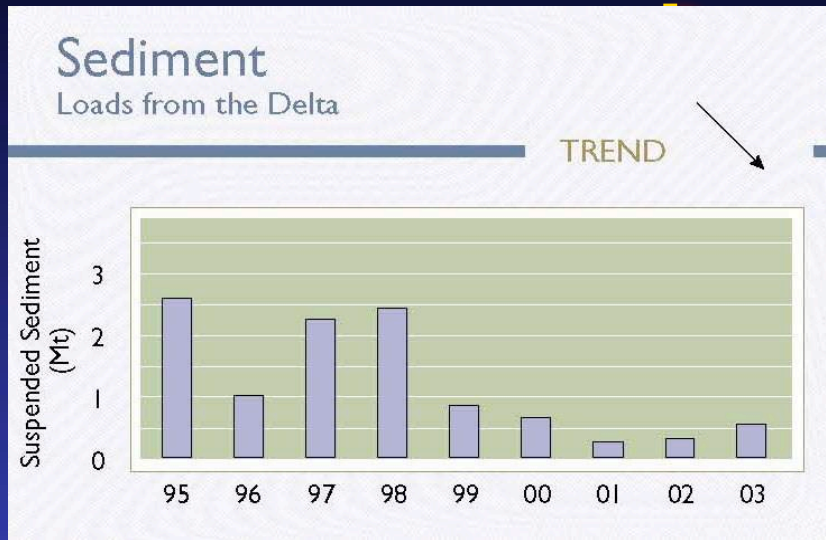
Source: Beutel et al., Mercury Technical Memo, 2004

T-Hg vs MeHg in Salt Ponds Sediment



Source: Miles et al., USGS - WERC, 2005

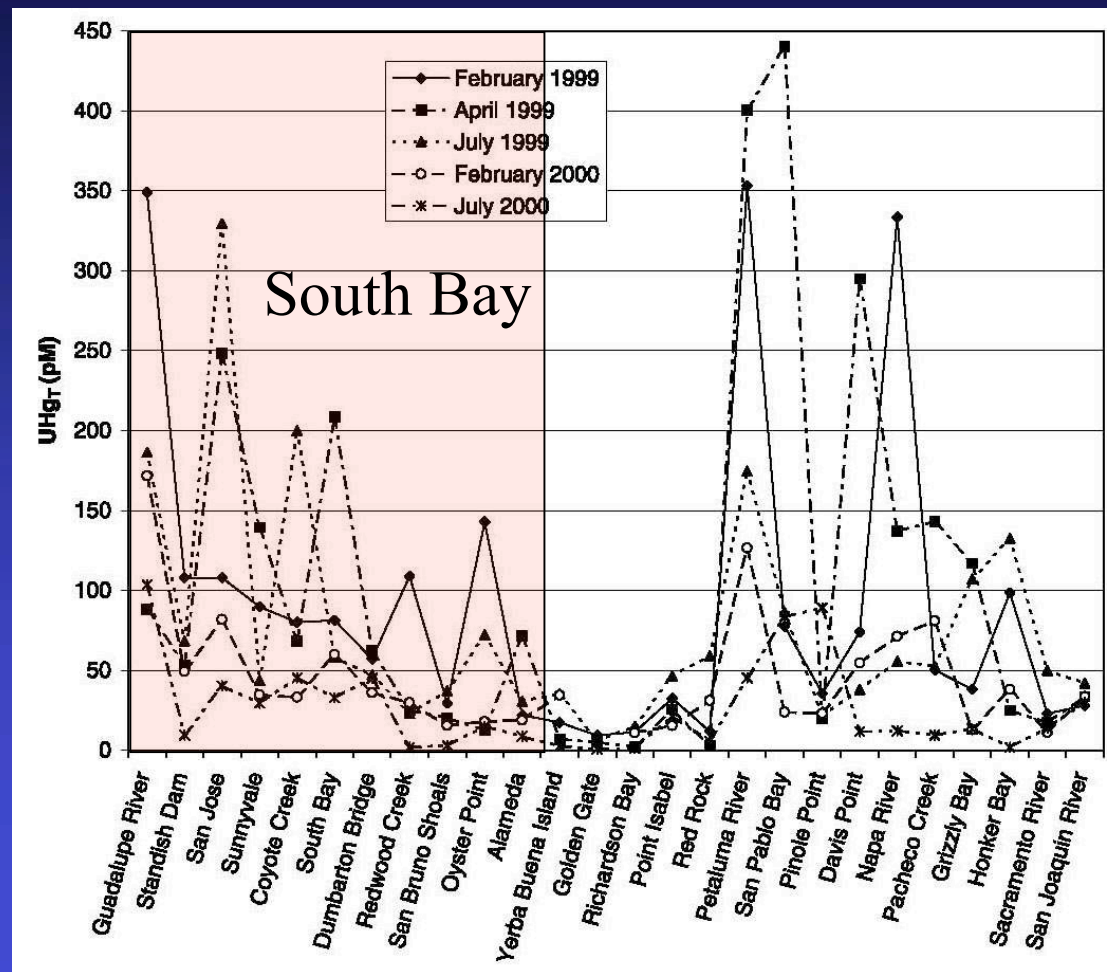
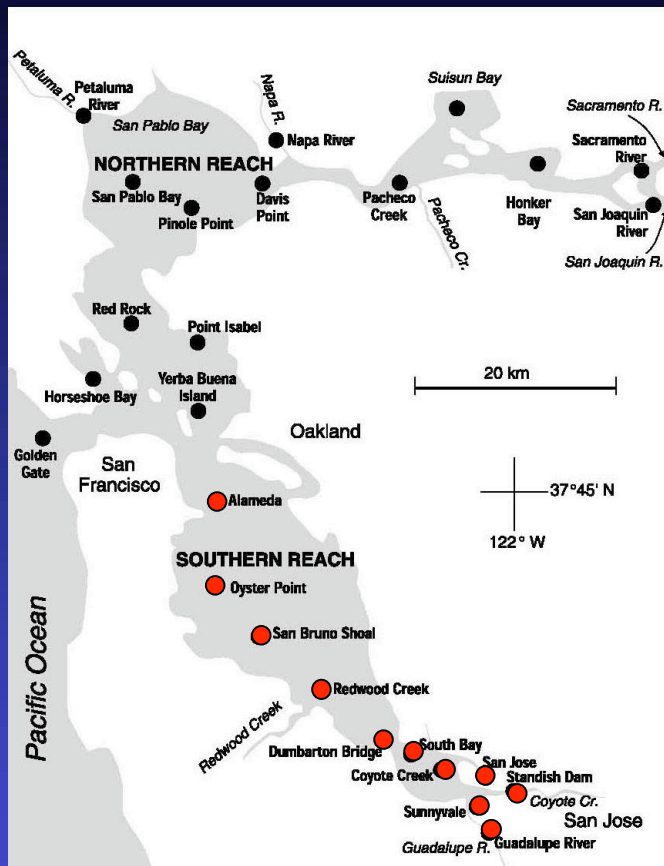
Sediment Associated Mercury



Source: Pulse of the Estuary 2005

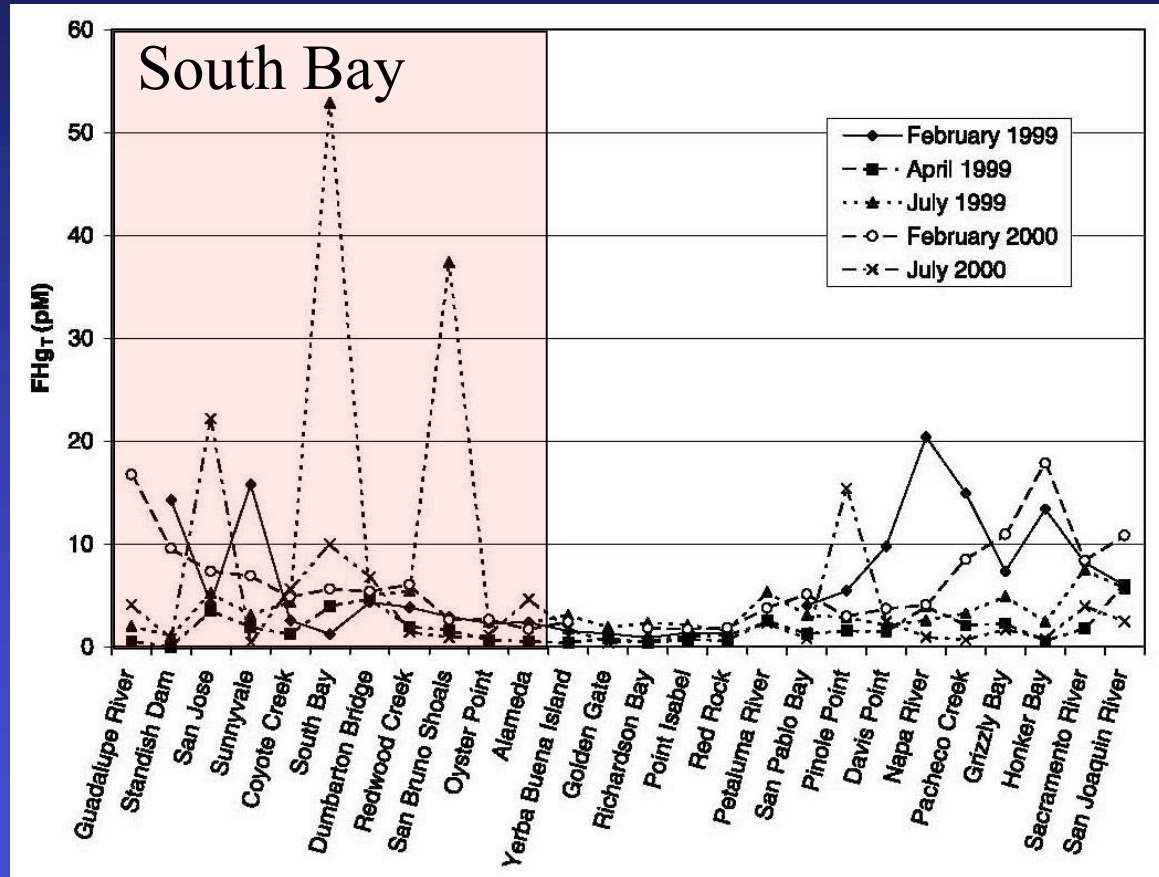
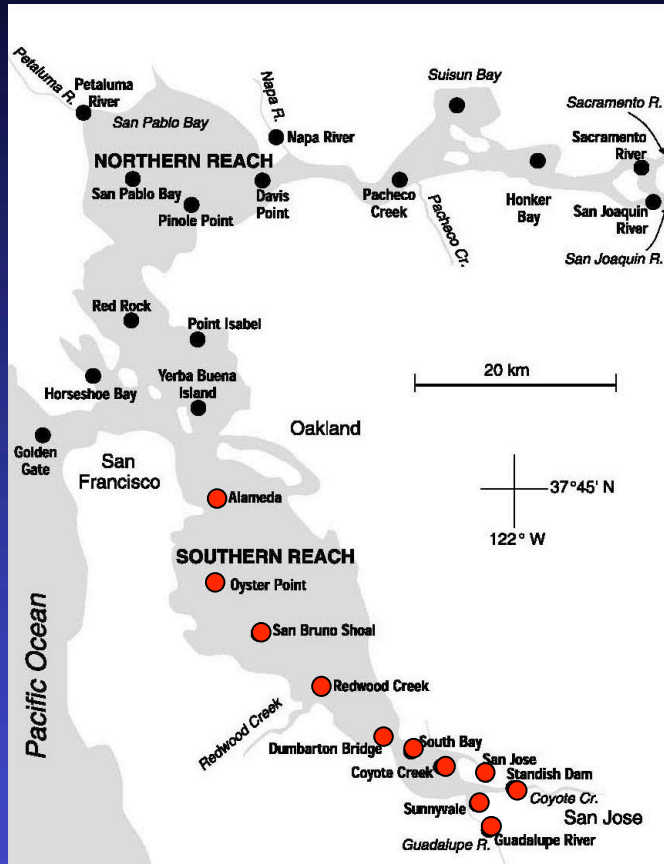
Data: RMP, USGS, SFEI (Lester McKee)

Unfiltered Total Hg in Water - SFB



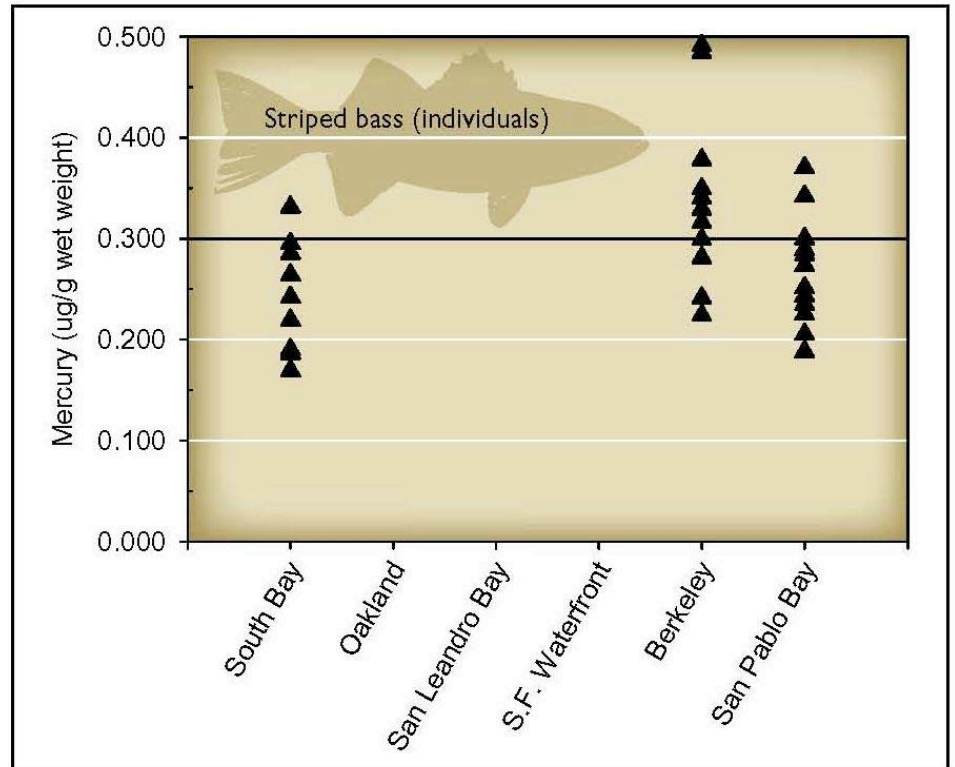
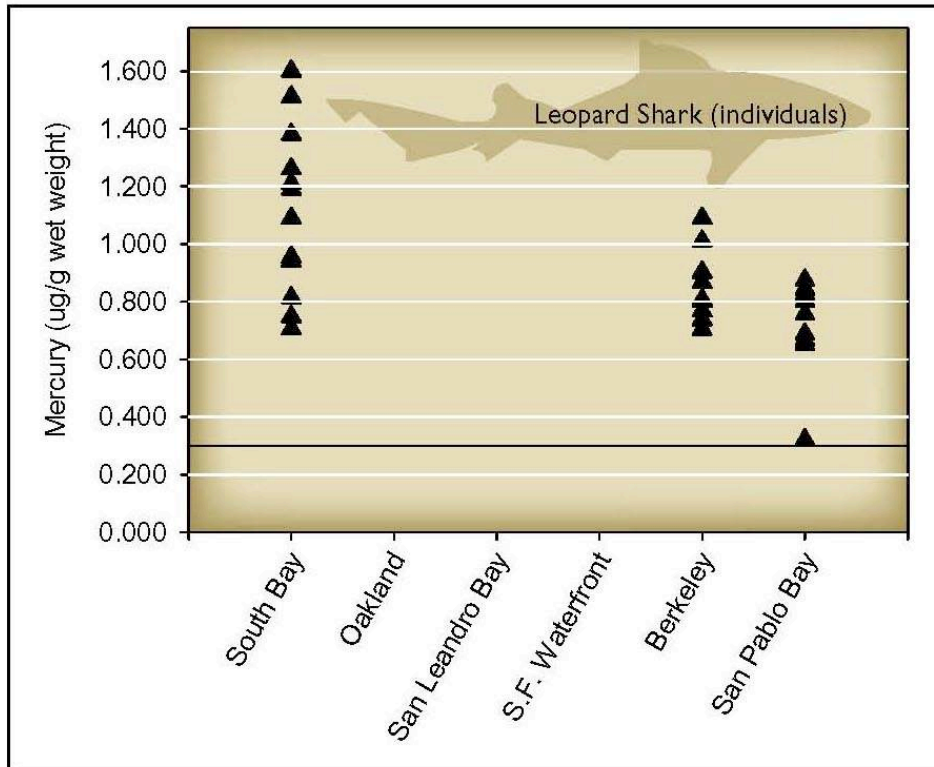
Source: Conaway et al. 2003, Marine Chemistry, v. 80 p. 199–225

Filtered Total Hg in Water - SFB



Source: Conaway et al. 2003, Marine Chemistry, v. 80 p. 199–225

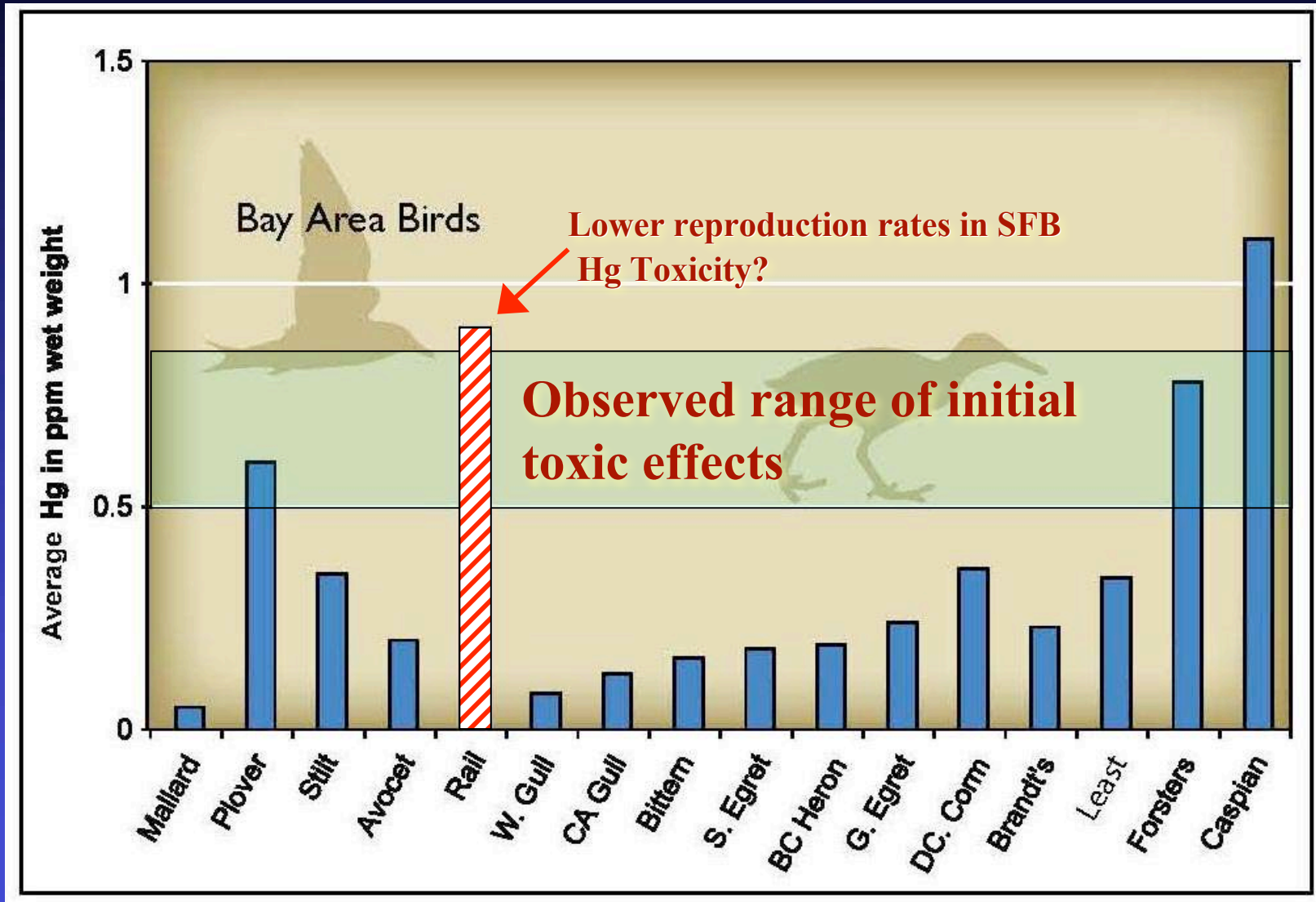
Mercury in Fish



Source: Pulse of the Estuary 2005

Data: RMP???

Mercury in Birds Eggs - SFB



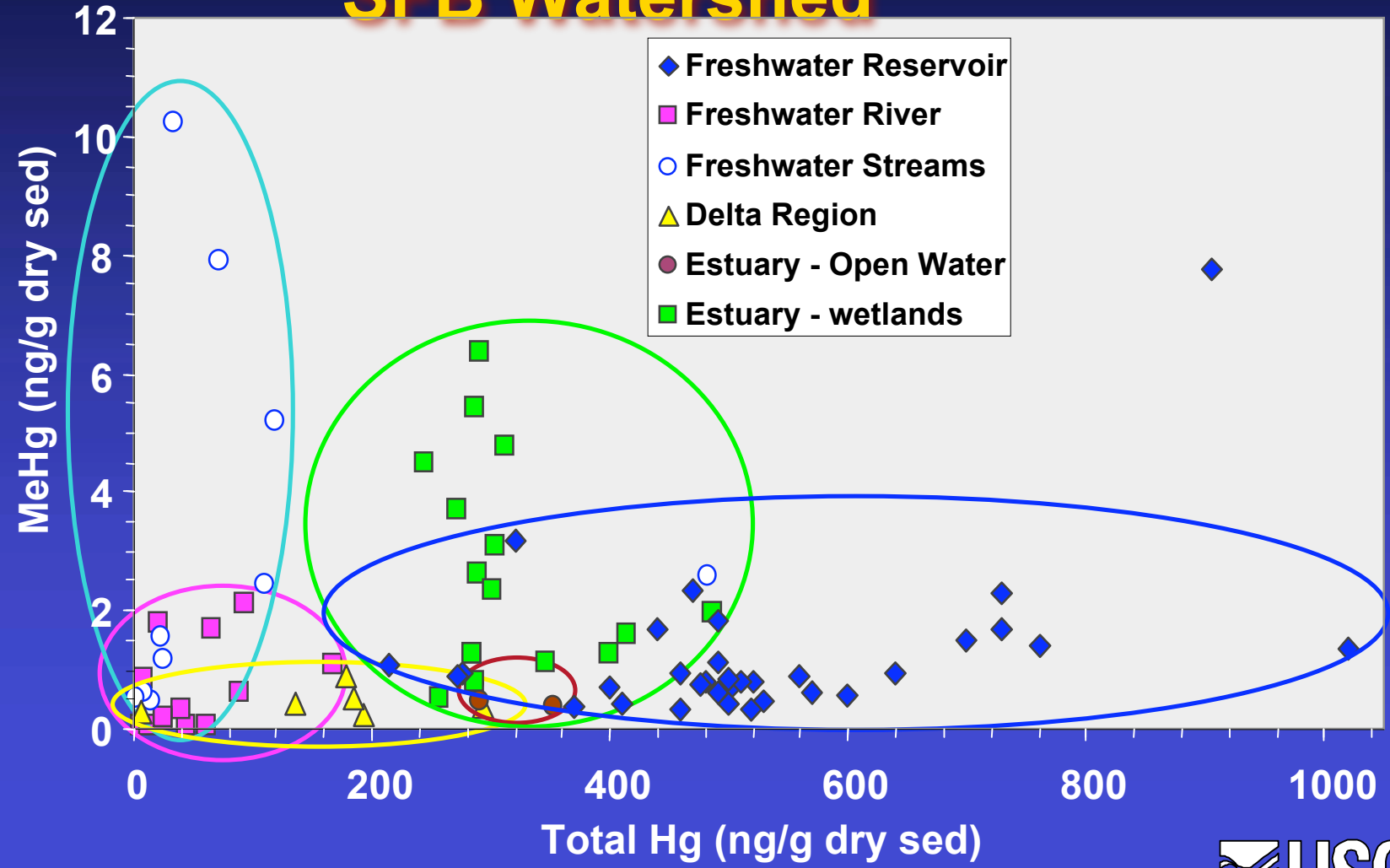
Source: Pulse of the Estuary 2005

Data: USFWS; Schwarzbach & Adelsbach, 2003

MeHg Production: Habitat Type Matters!



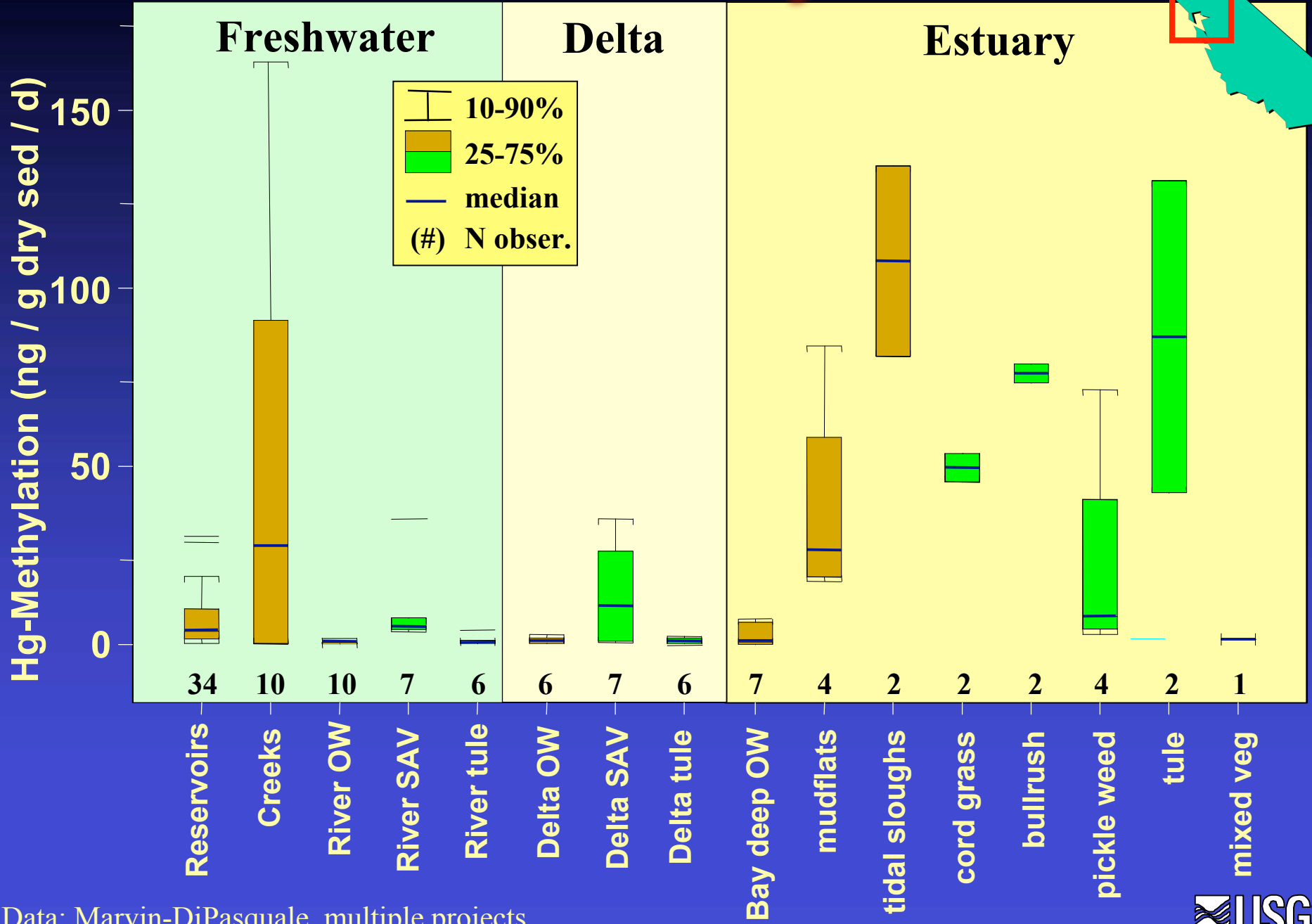
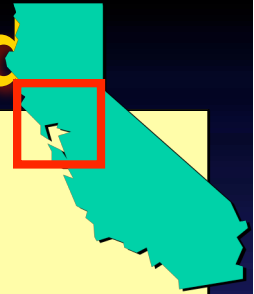
SFB Watershed



Data: Marvin-DiPasquale, multiple projects



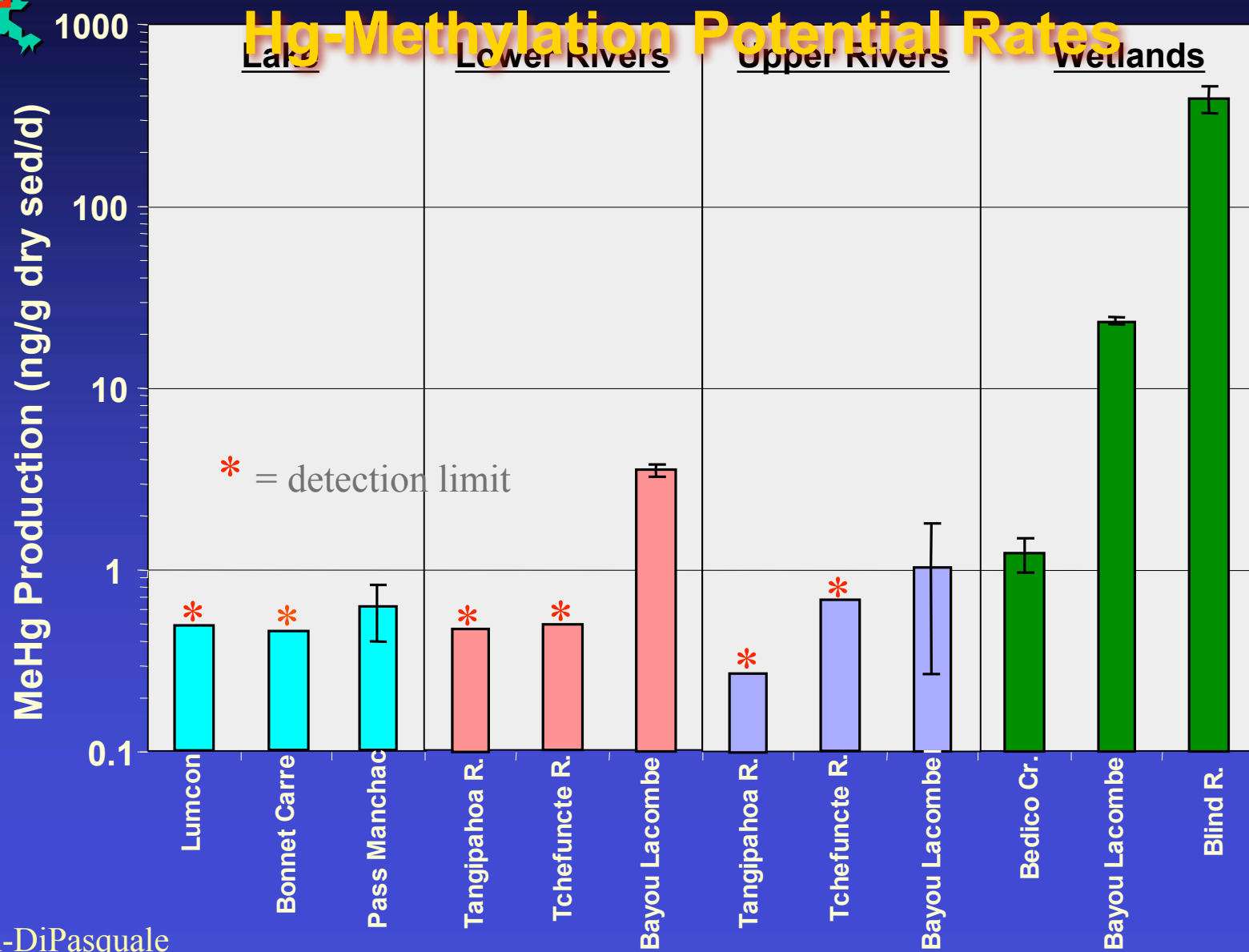
San Francisco Bay Watershed



Data: Marvin-DiPasquale, multiple projects



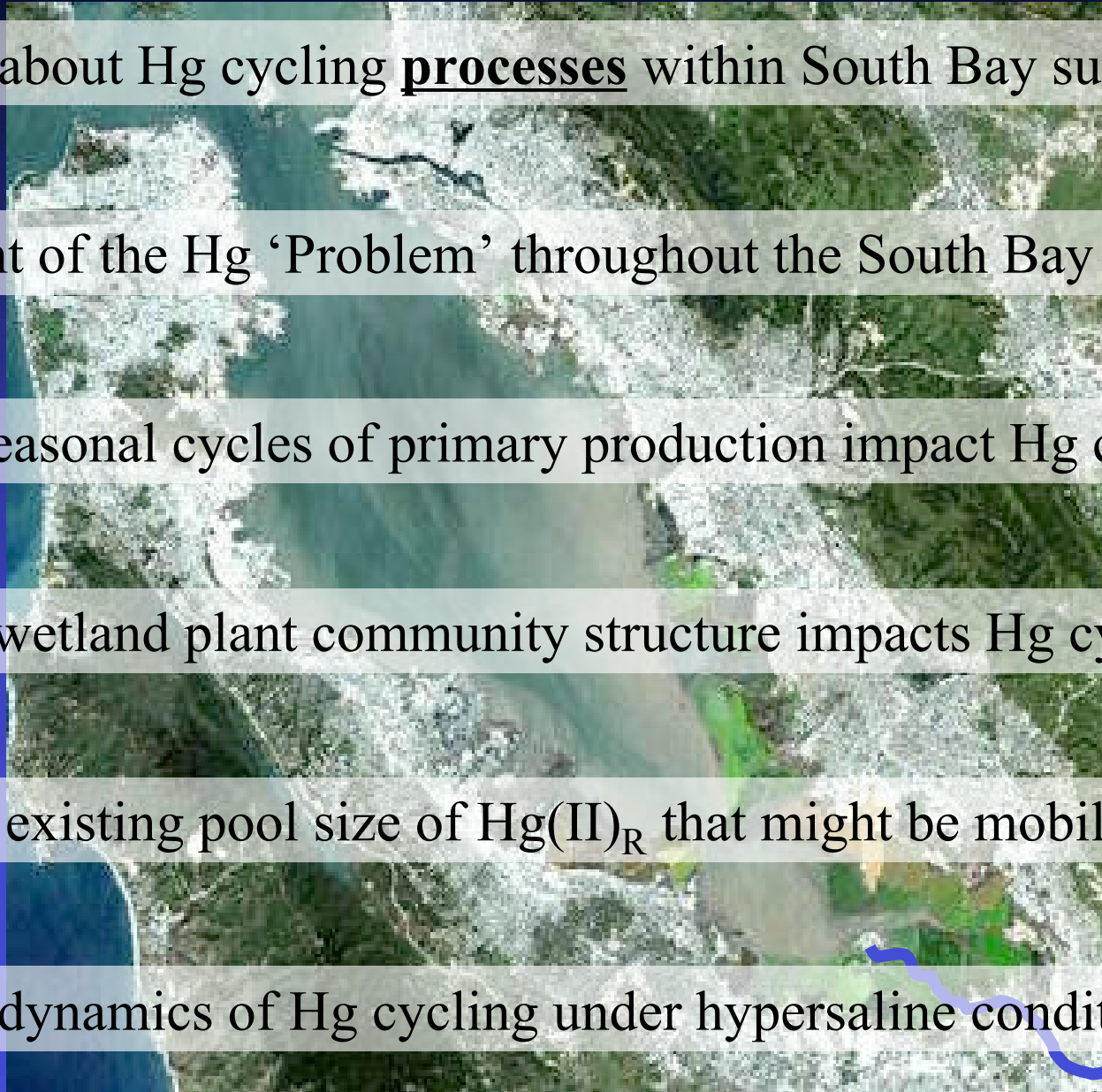
Lake Pontchartrain Watershed (LA), August '03



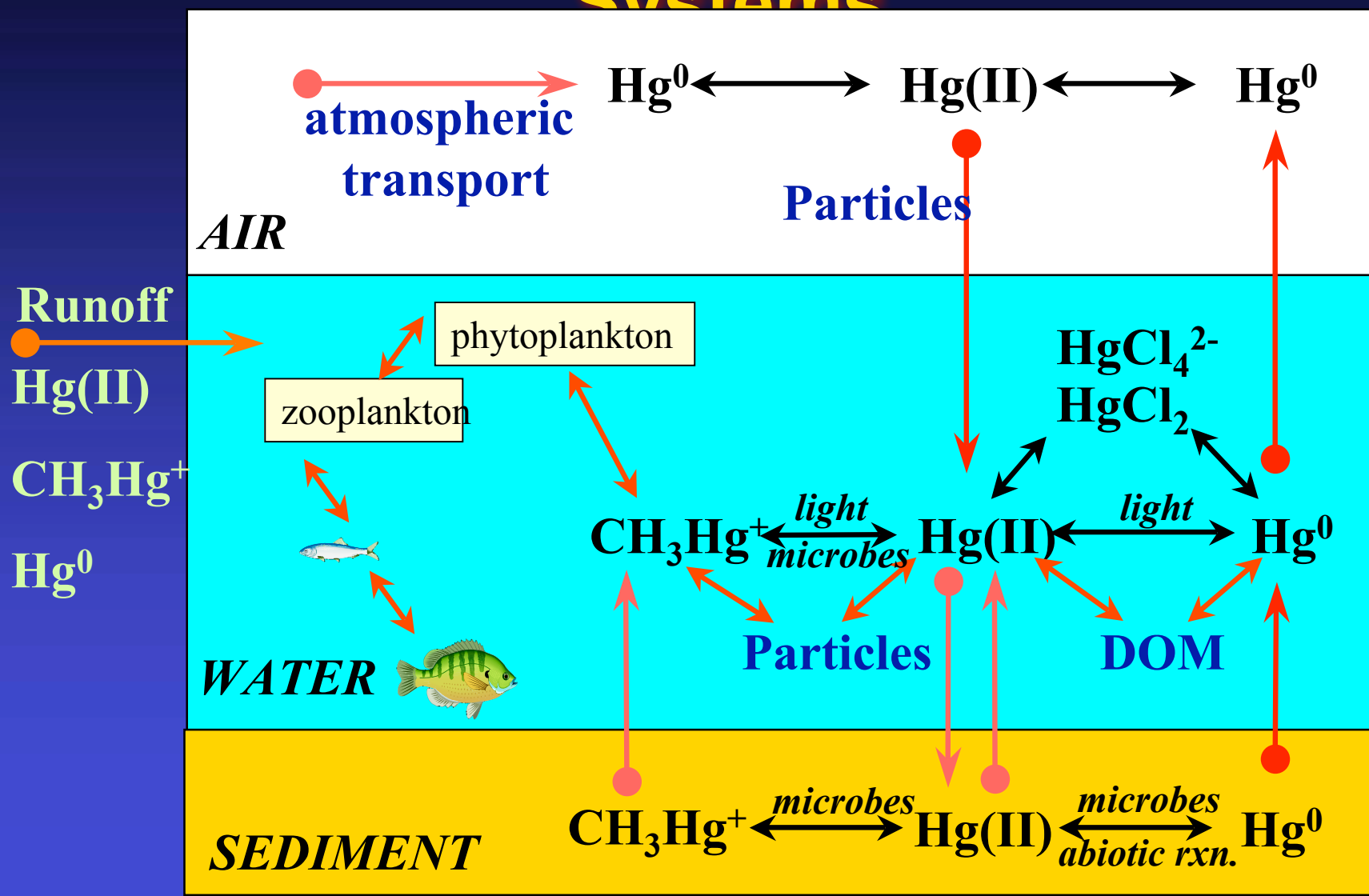
Data: Marvin-DiPasquale

What Don't We Know?

- ✓ Specifics about Hg cycling processes within South Bay sub-habitats
- ✓ The extent of the Hg 'Problem' throughout the South Bay food web
- ✓ How seasonal cycles of primary production impact Hg cycling
- ✓ How wetland plant community structure impacts Hg cycling
- ✓ The existing pool size of Hg(II)_R that might be mobilized
- ✓ The dynamics of Hg cycling under hypersaline conditions



The Mercury Cycle in Aquatic Systems



What Controls MeHg Production?

In simplest terms:

$$\text{Gross MeHg Production} \\ - \text{Gross MeHg Degradation}^*$$

Net MeHg Production

What Controls NET MeHg Production?

MICROBIAL ACTIVITY

*Hg(II)-Methylating &
MeHg-Degrading*

- community composition
- electron donors (org-C)
- electron acceptors
(e.g. Fe^{3+} , SO_4^{2-} , NO_3^-)
- pH and redox (E_h)
- temperature

MERCURY AVAILABILITY

Reactive-Hg(II) and MeHg

- amount of total-Hg or MeHg
- complexation w/ DOM
- binding to particles
- pH and redox (E_h)
- sulfur and iron chemistry
- salinity (Cl^-)
- chemical form: (HgS , HgS_2^{2-} ,
 HgCl_2 , HgCl_4^{2-} , etc...)

Hot Spot or Not?: $MP = K \times Hg(II)_R$

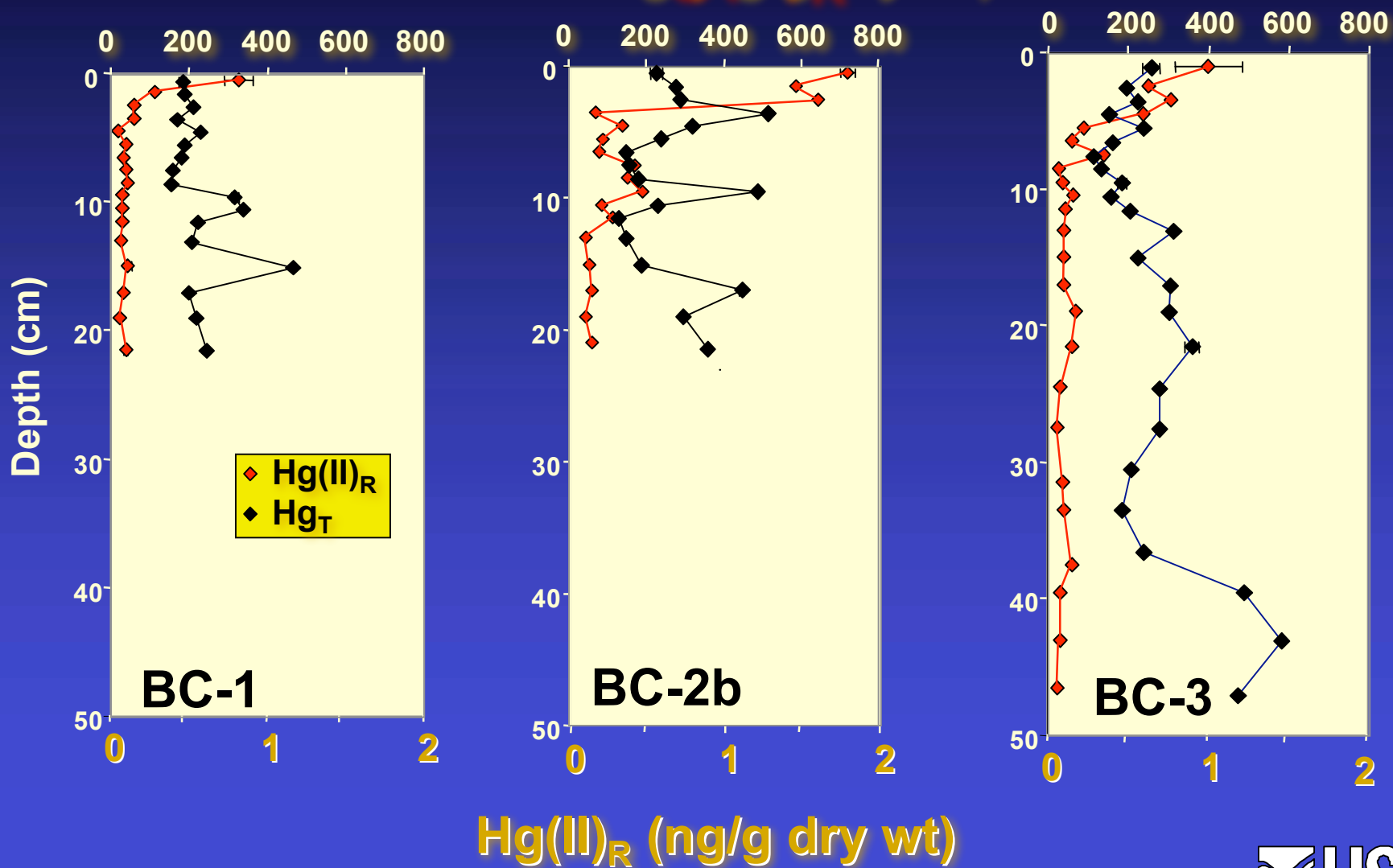
Reactive Hg(II) Concentration

Hg(II)-Methylating
Bacteria Activity (K)

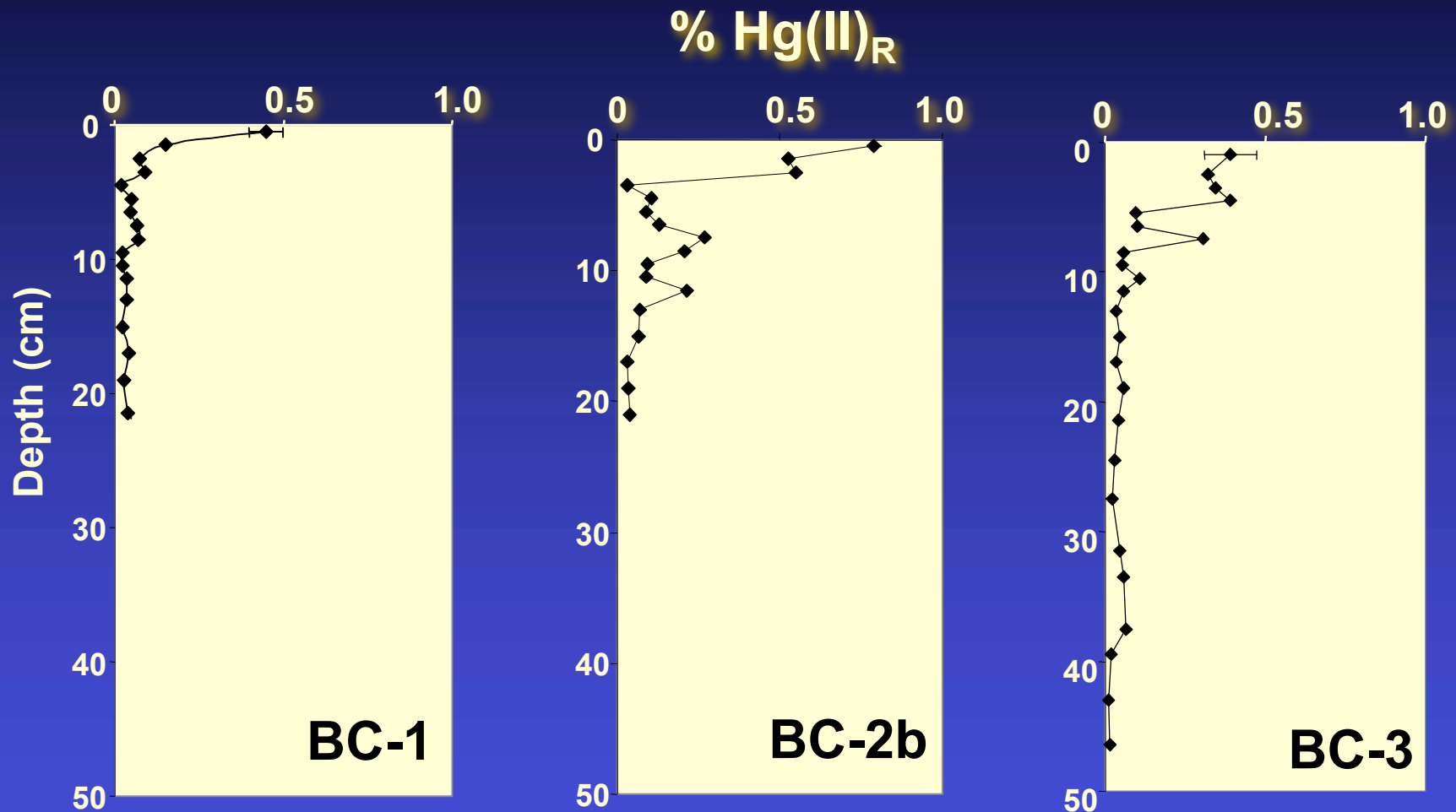
	Low	High
Low	<u>Low</u> MeHg Production	<u>Moderate</u> MeHg Production
High	<u>Moderate</u> MeHg Production	<u>High</u> MeHg Production

San Pablo Bay (Sept. '04) – Hg_T &

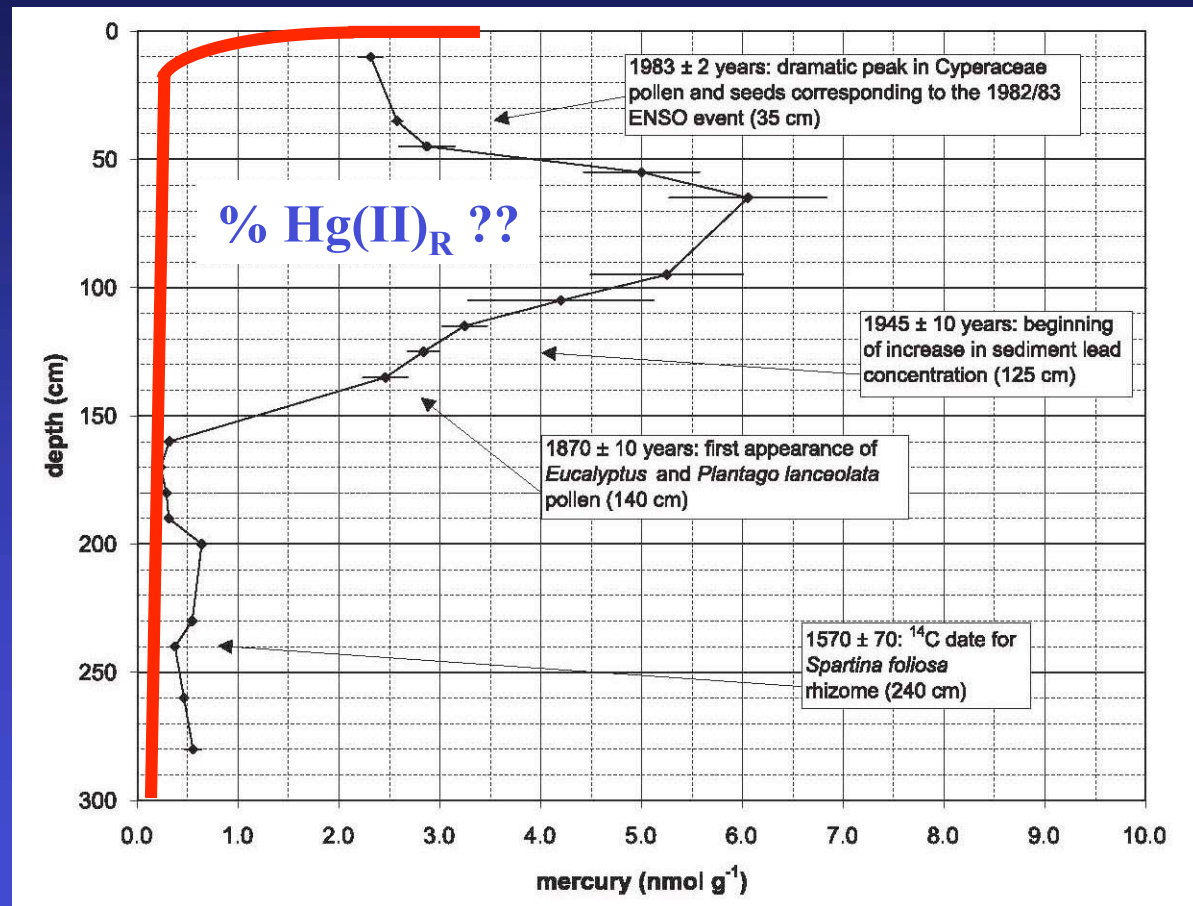
Hg(II)_R Total Hg (ng/g dry wt)



San Pablo Bay (Sept. '04) – Percent Hg(II)_R

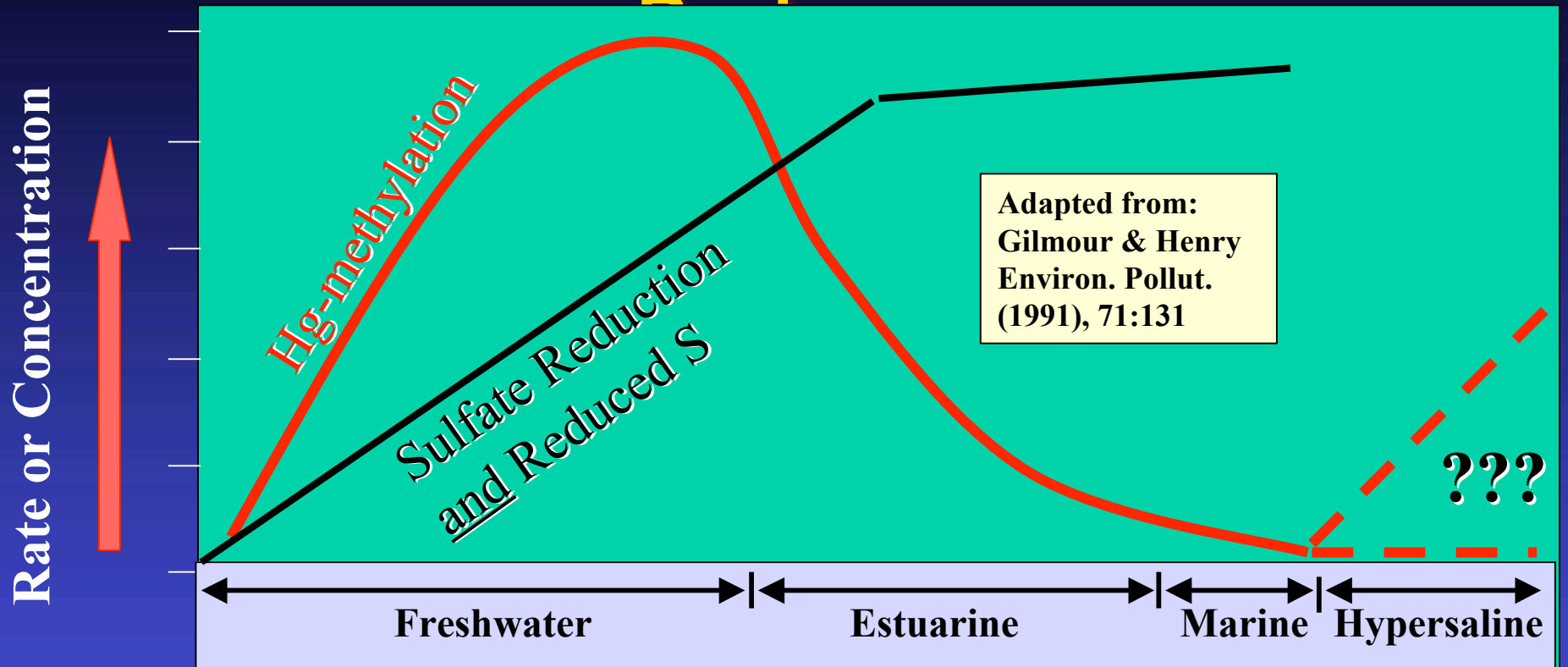


Mercury in Wetland Sediment Triangle Marsh – Dated Core



Source: Conaway et al. 2004, Marine Chemistry, v. 90 p. 175–184

Hg-Methylation Activity Across Salinity



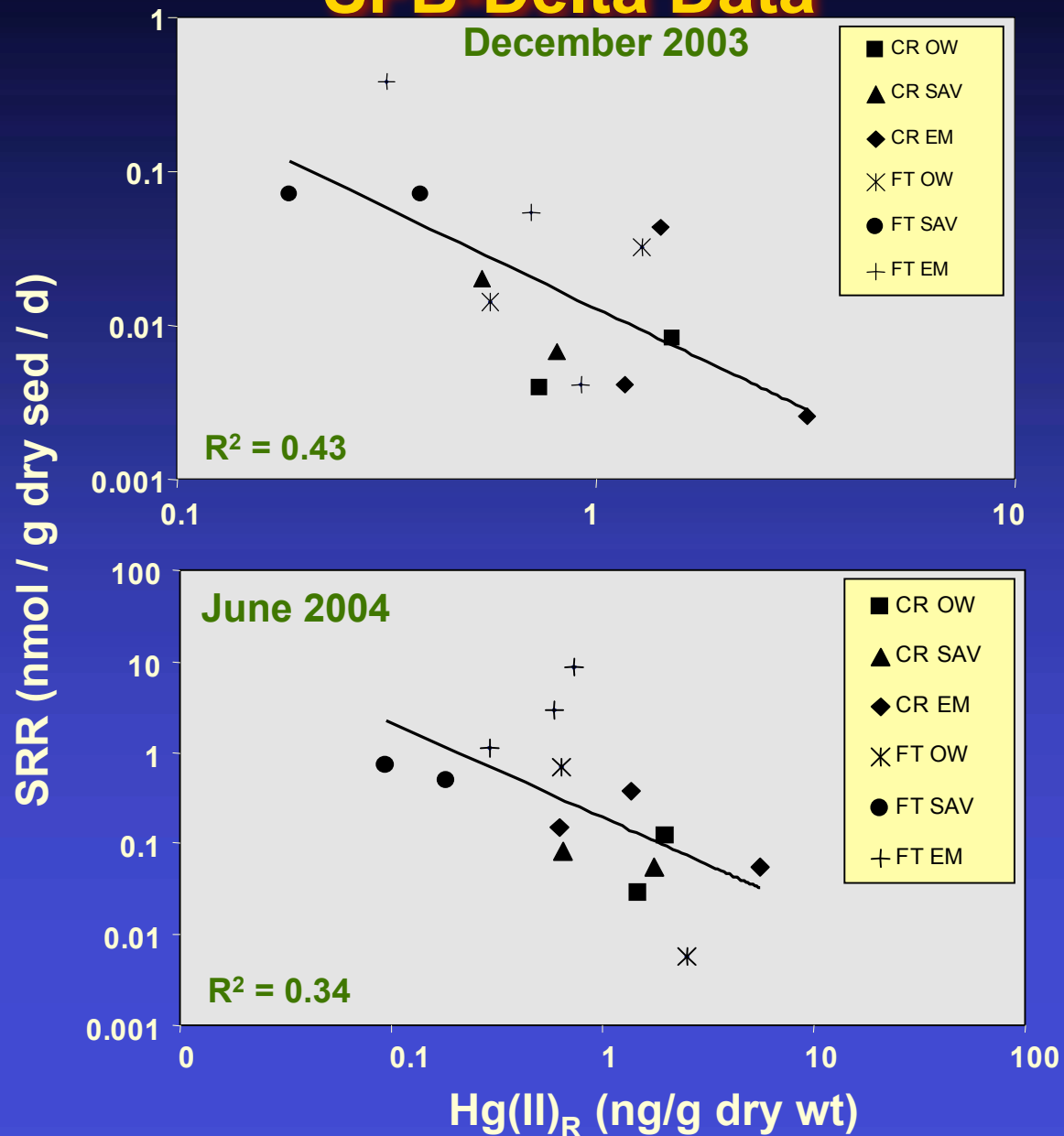
Low SO_4^{2-} & Cl^-
 Low Sulfate Reduction
 Low Reduced-S
 Dominant: HgS & HgCl_2
 (neutral)



High SO_4^{2-} & Cl^-
 High Sulfate Reduction
 High Reduced-S
 Dominant: HgS_2^{2-} &
 HgCl_4^{2-} (charged)

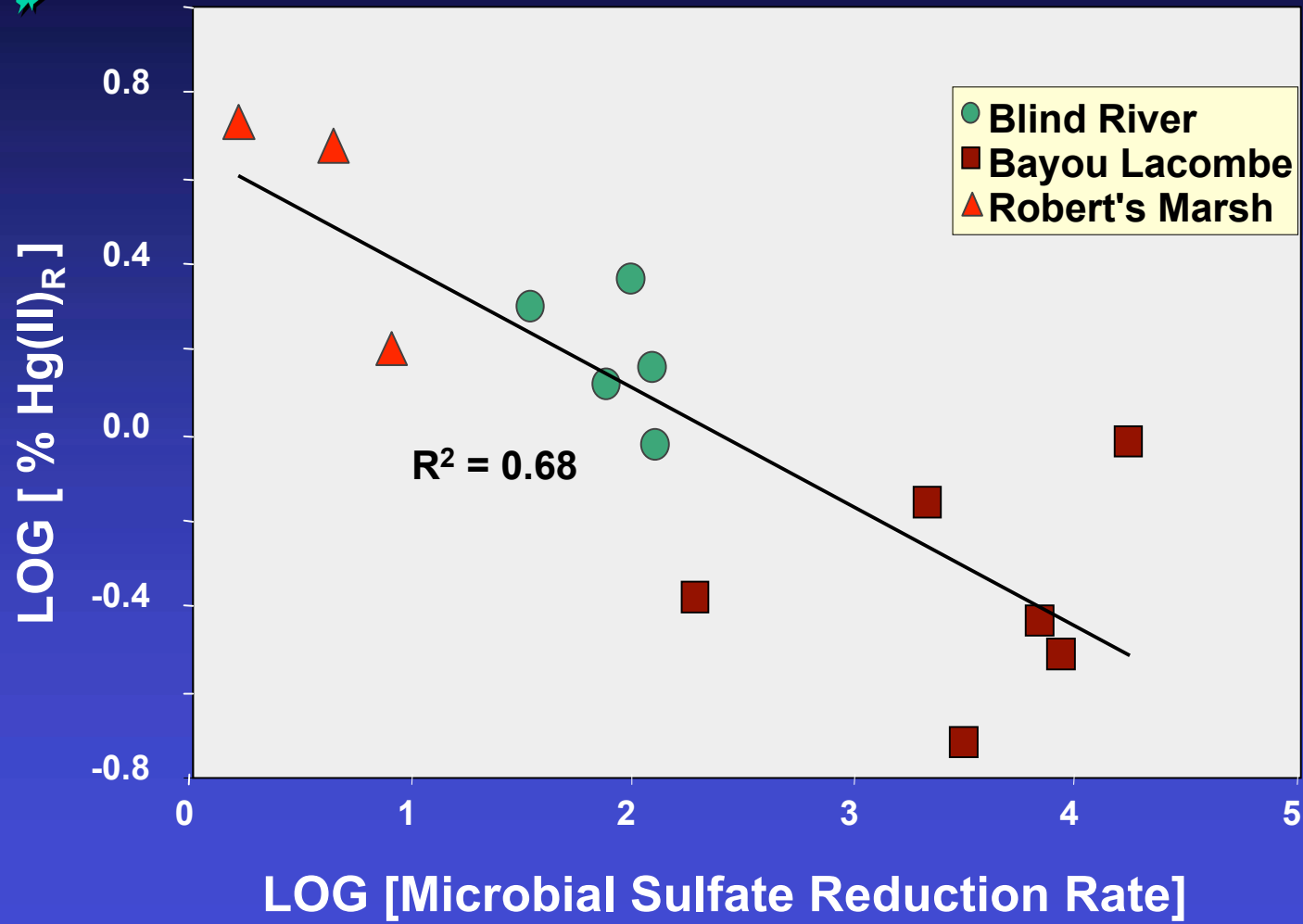
Sulfate Reduction: A Control on Hg(II)_R ?

SFB-Delta Data



Sulfate Reduction: A Control on Hg(II)_R ?

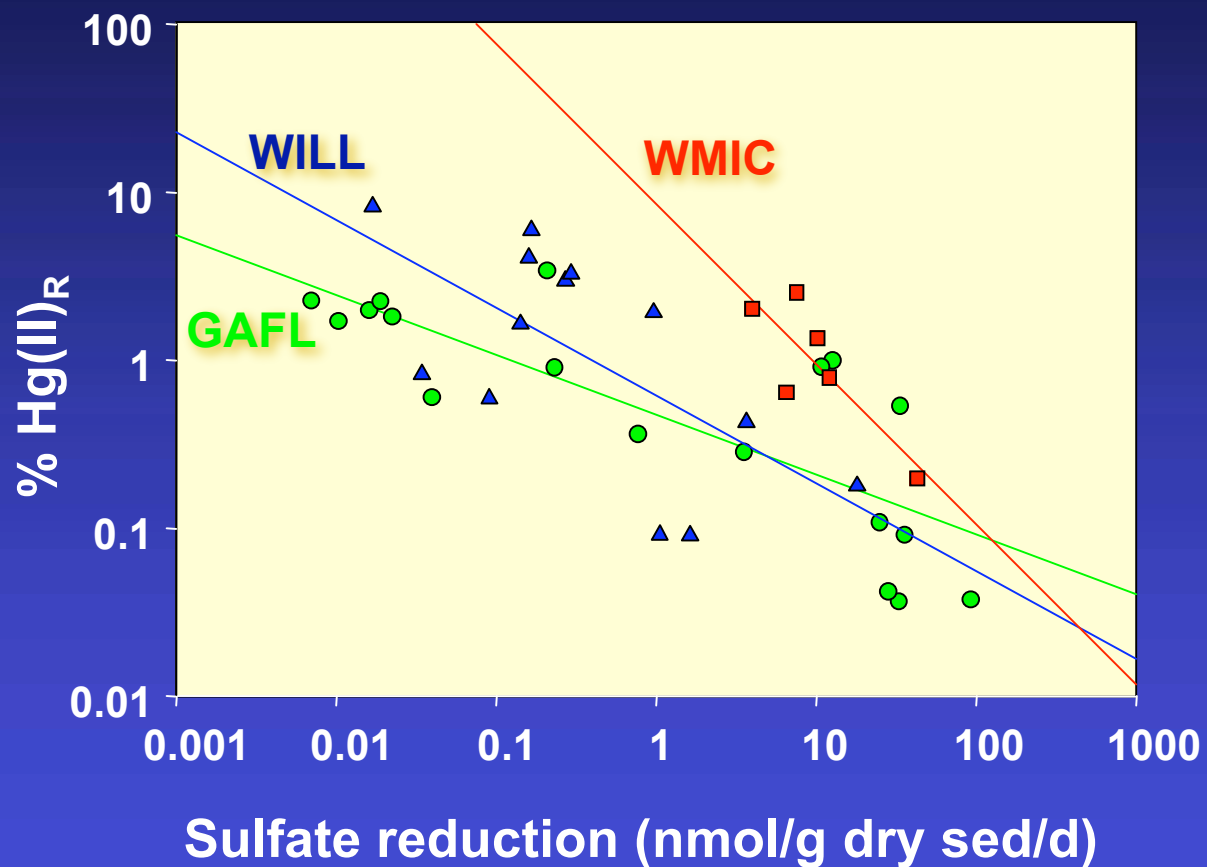
Louisiana Wetlands, April '04, surface 0-2 cm



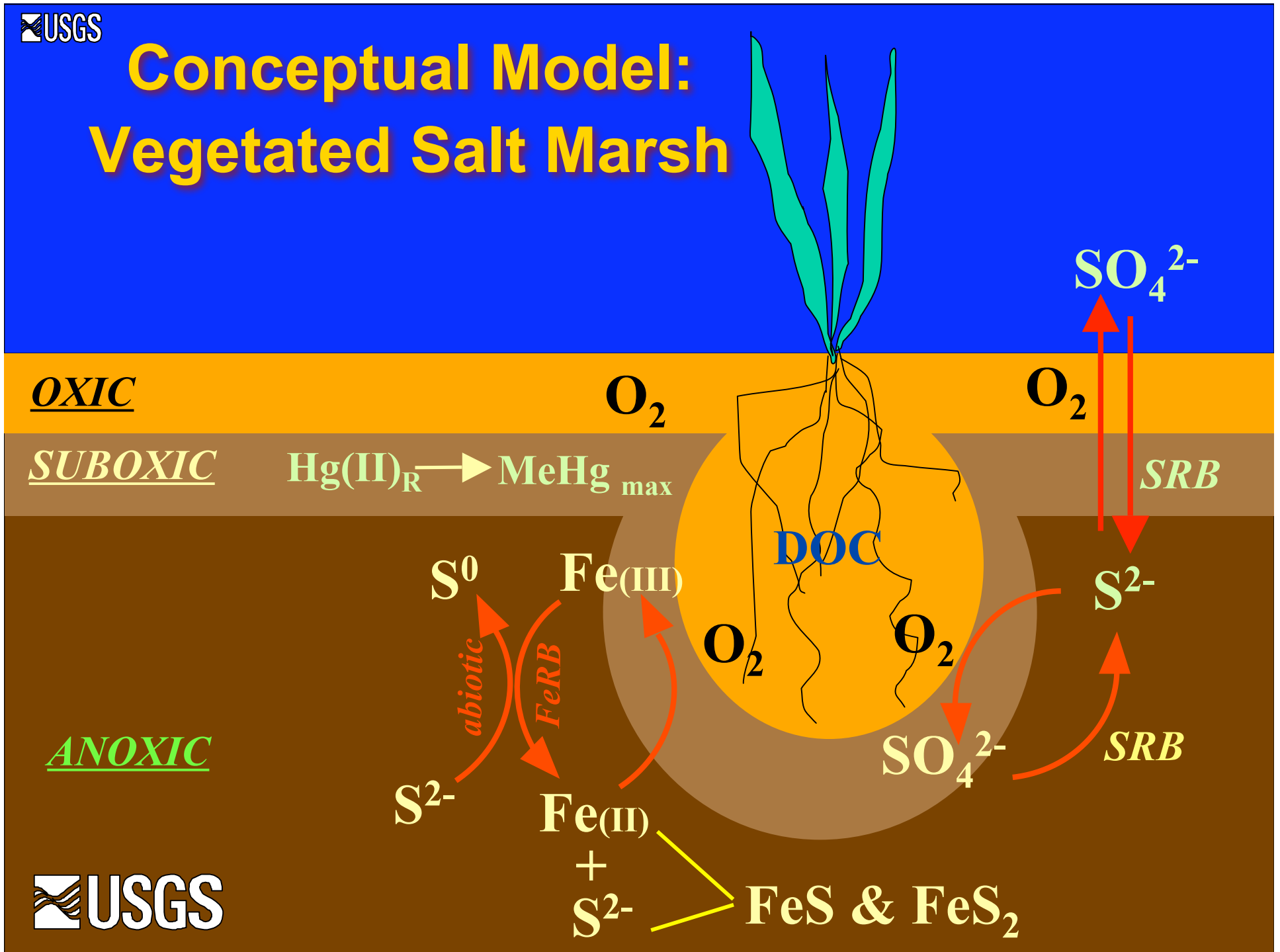
Sulfate Reduction: A Control on Hg(II)_R ?

Stream & River Data from 3 National Regions

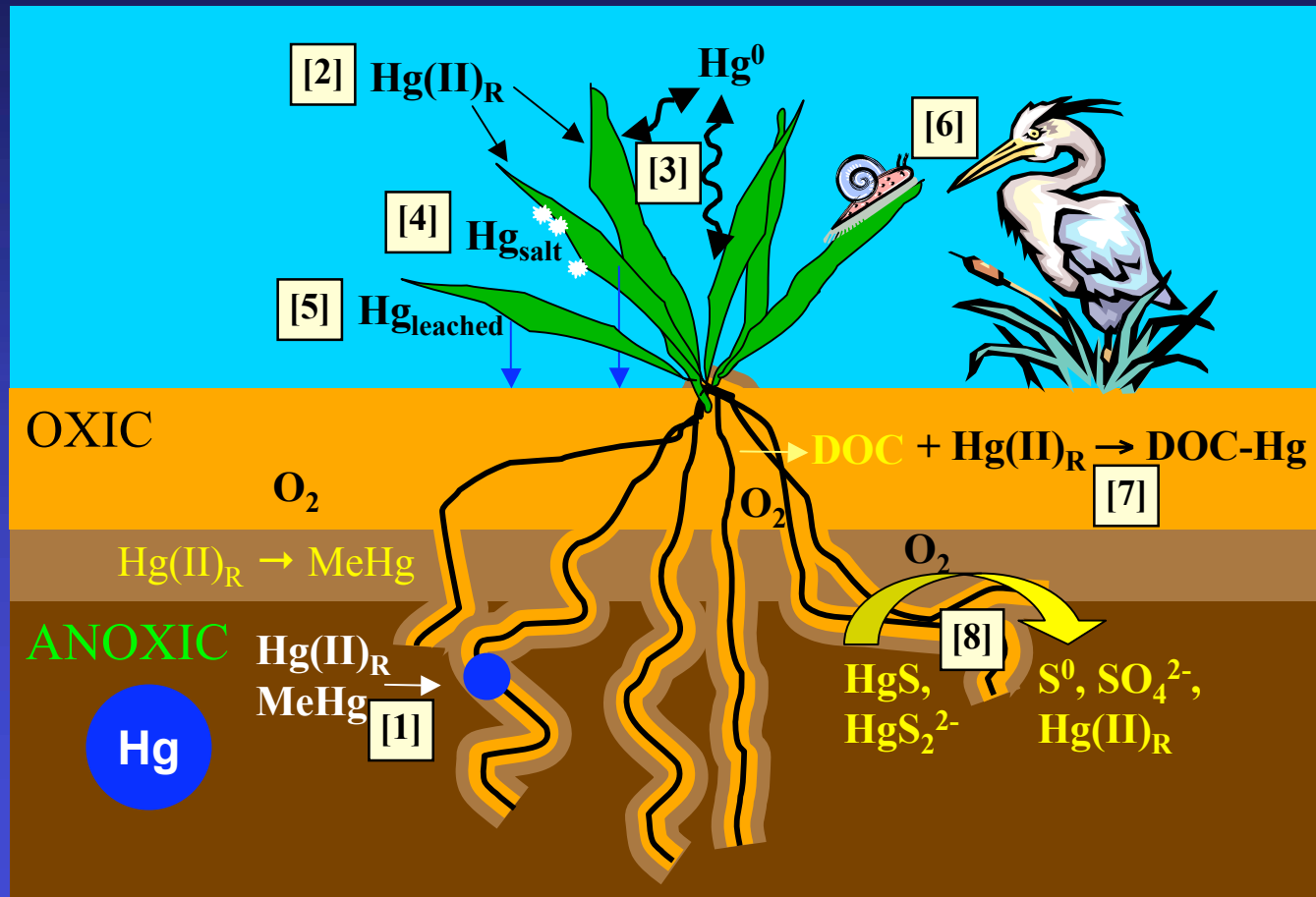
USGS-NWAQA Mercury Topical Study



Conceptual Model: Vegetated Salt Marsh

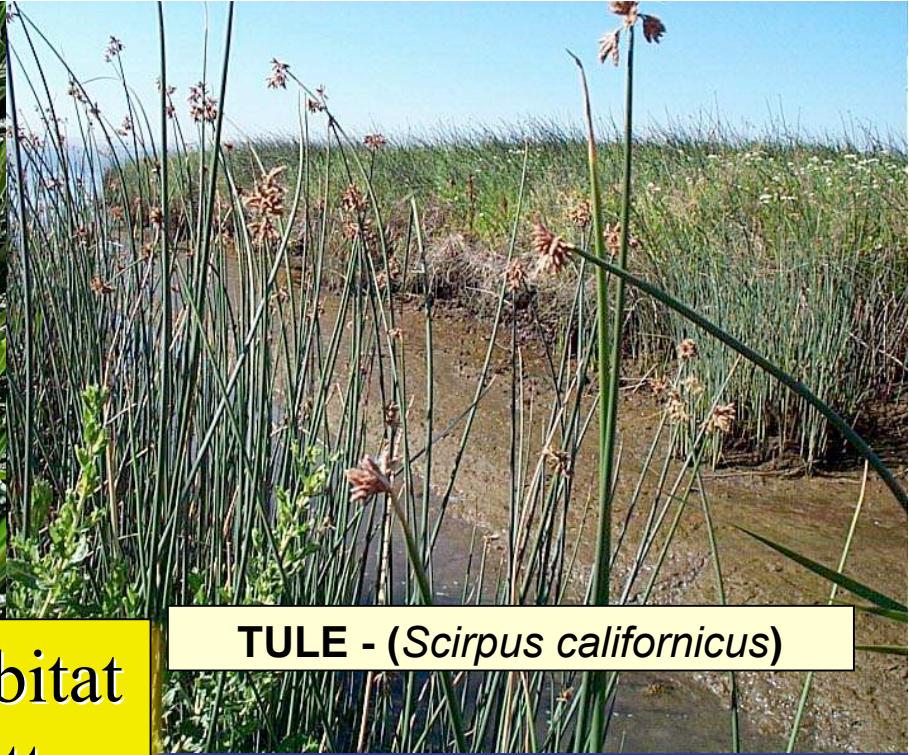


Wetland plants can alter Hg pools and fluxes in MANY ways.





CORD GRASS - (*Spartina foliosa*)



TULE - (*Scirpus californicus*)

Habitat Matters



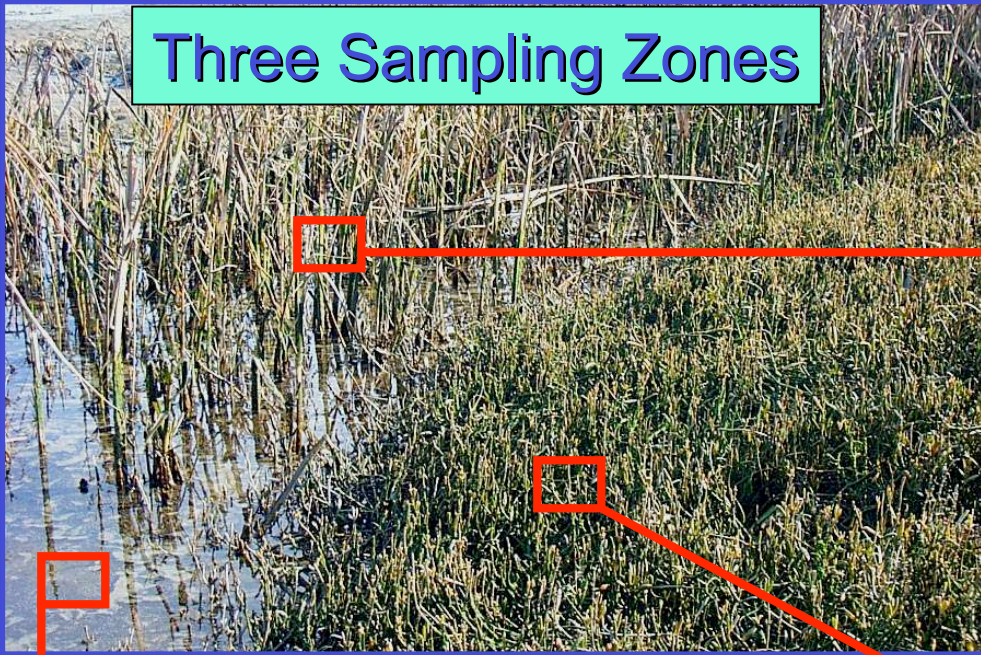
PICKLEWEED (*Salicornia virginica*)

Fe(III) →



Steven's Creek Marsh- South

Three Sampling Zones



bullrush root zone



non-vegetated mudflat

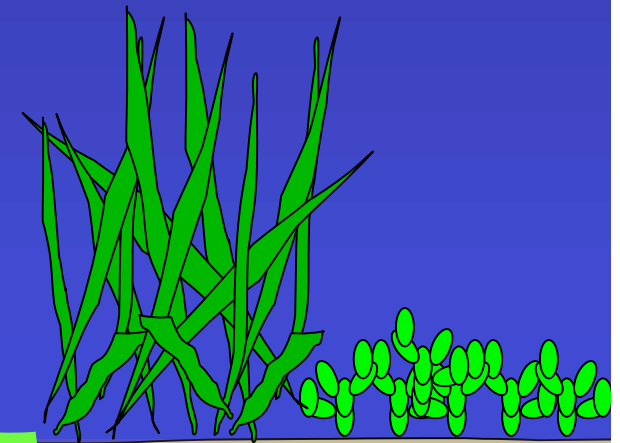
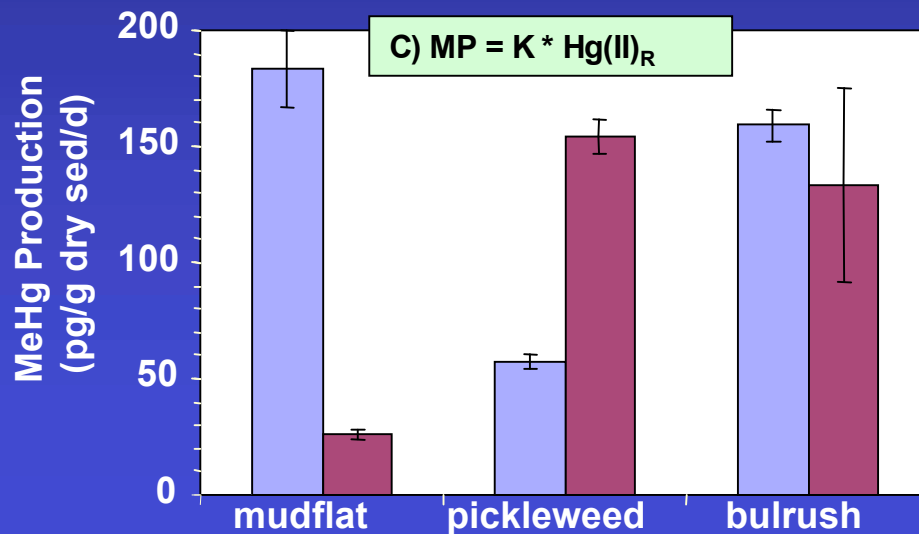
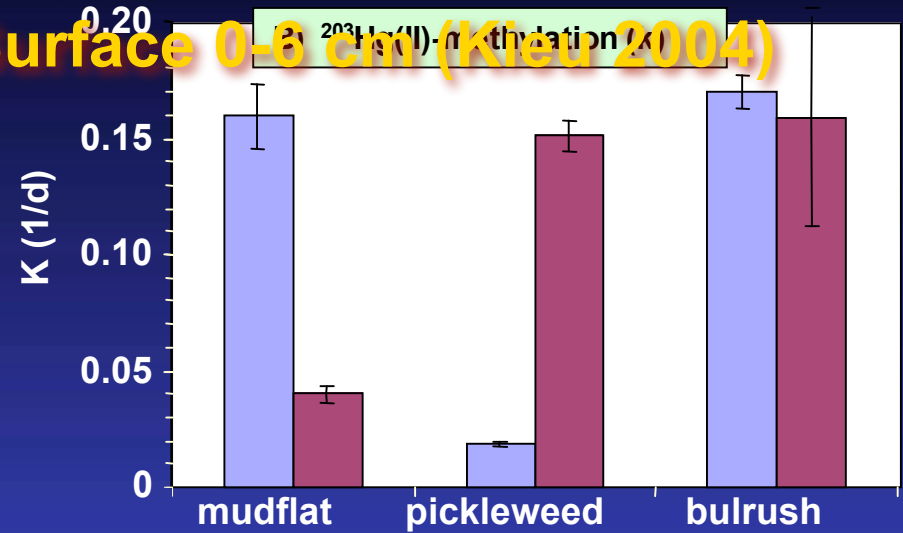
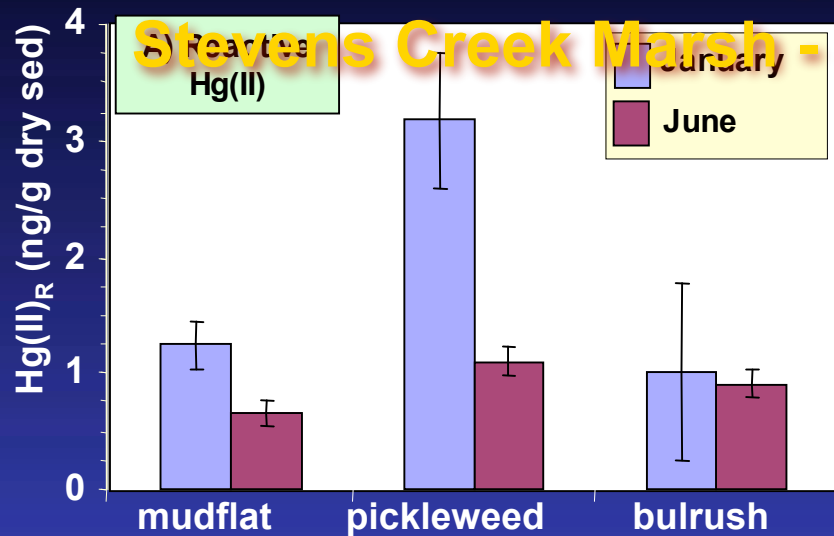


pickleweed root zone



Vegetation Affects Hg Cycling Over Small Spatial Scales

Stevens Creek Marsh - Surface 0-8 cm (Kieu 2004)



Mudflat Bulrush Pickleweed

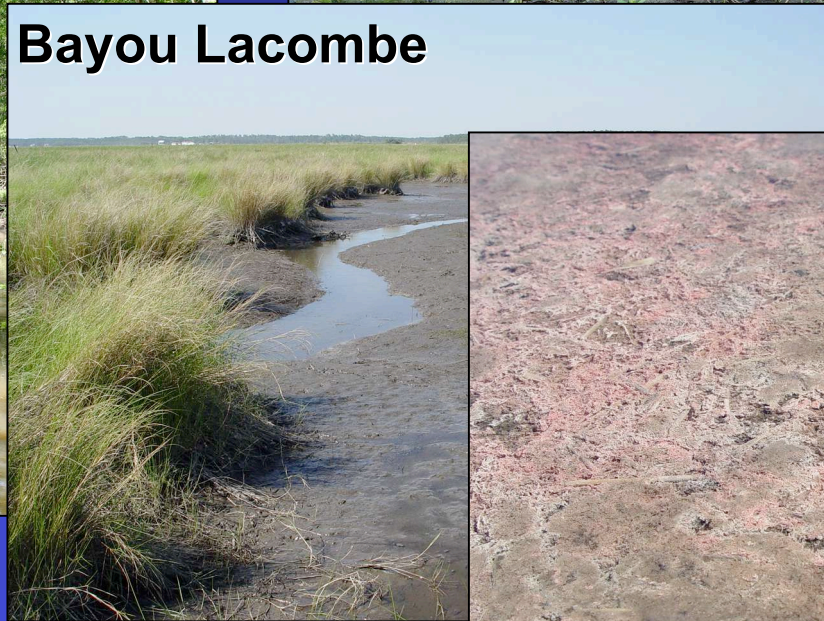
Louisiana Wetlands (April '04)



Blind River



Robert's Marsh
(Skull Creek)



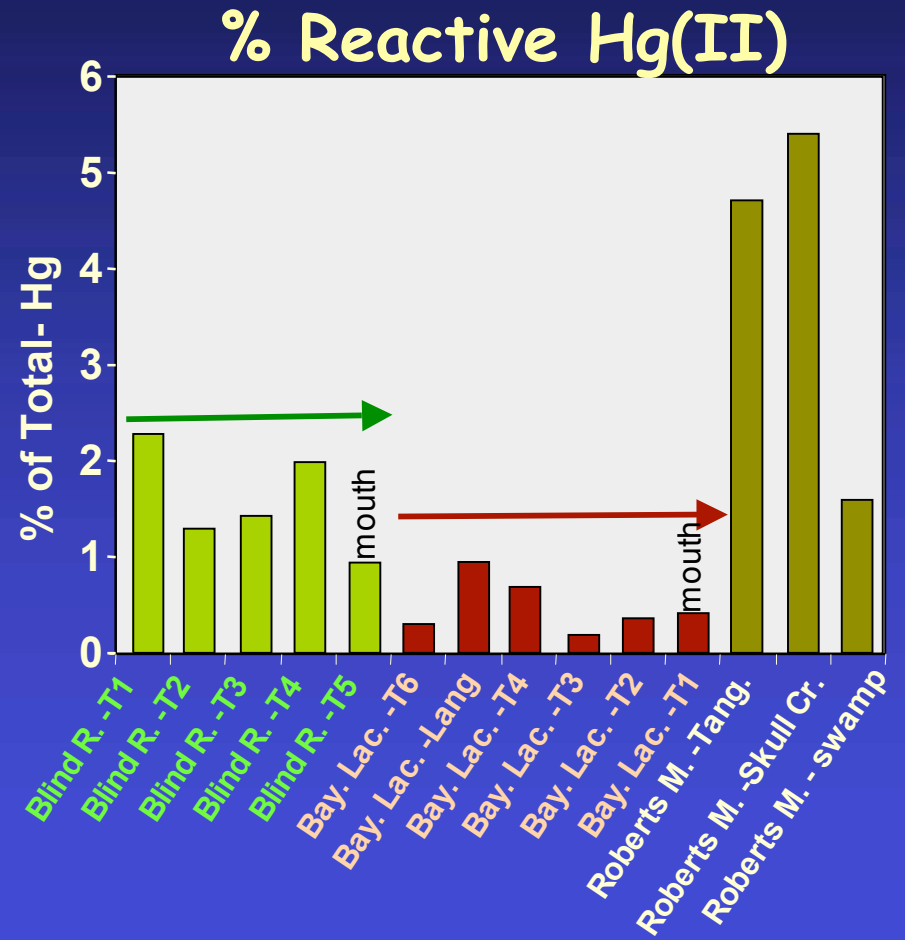
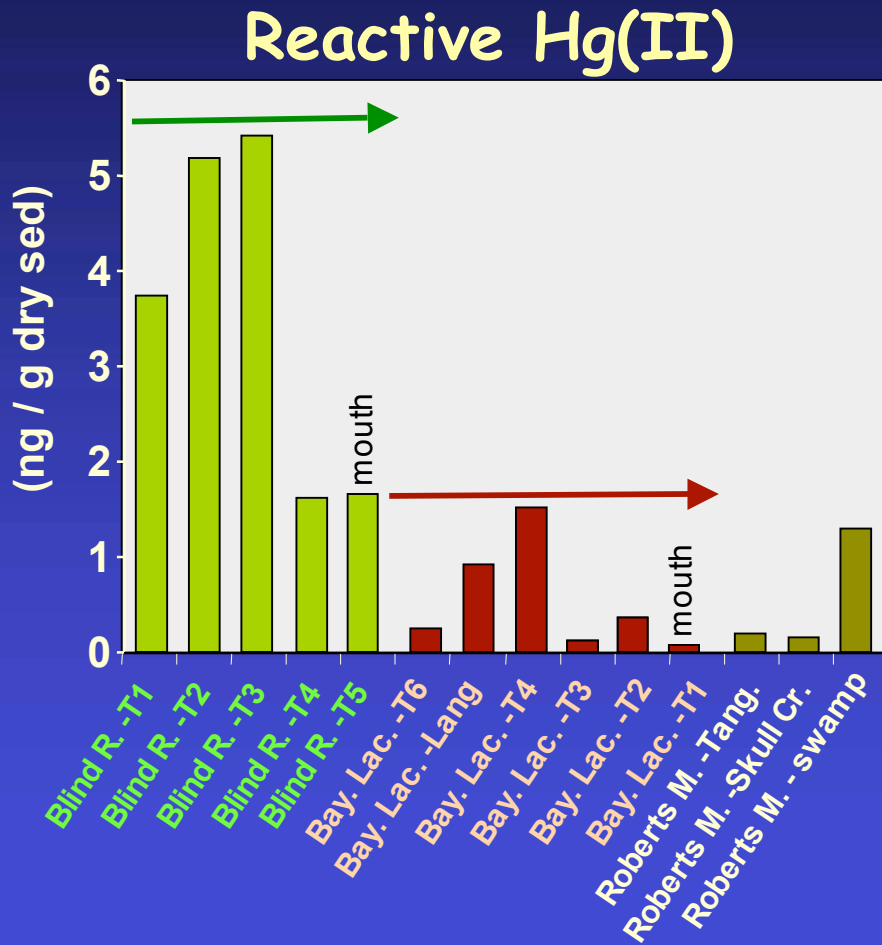
Bayou Lacombe



Langniappe!



Reactive Hg(II): Louisiana Wetlands, April '04 (surface 0-2 cm interval)

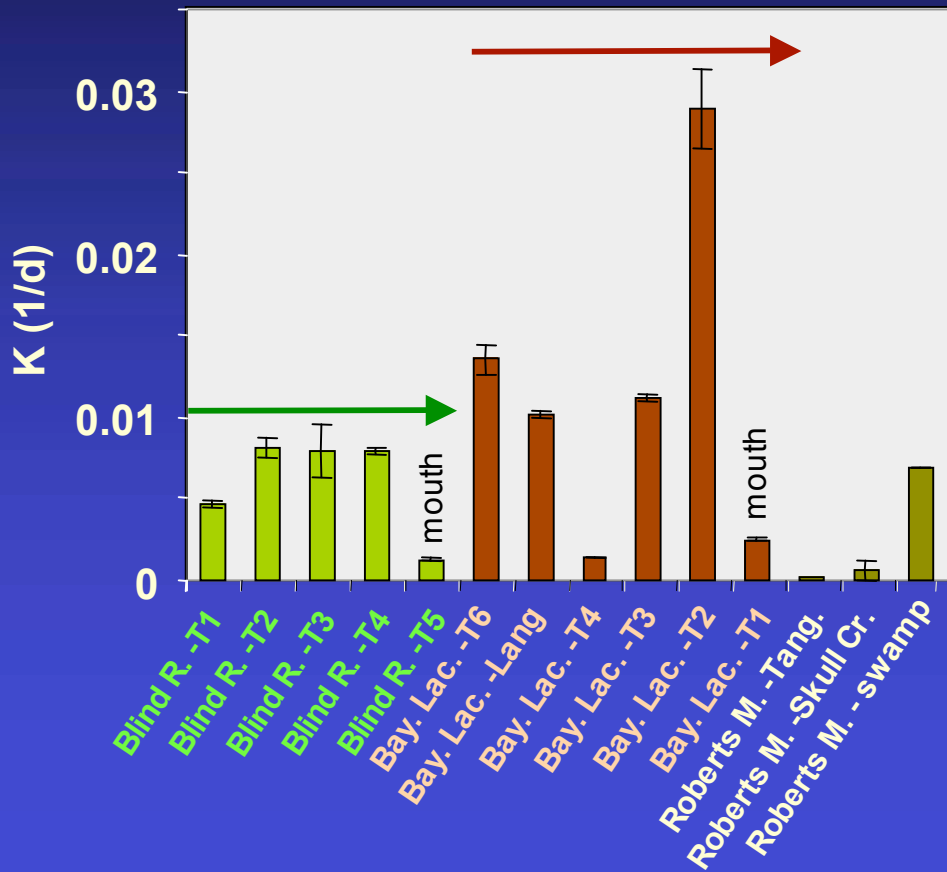




MeHg Production: Louisiana Wetlands, April '04

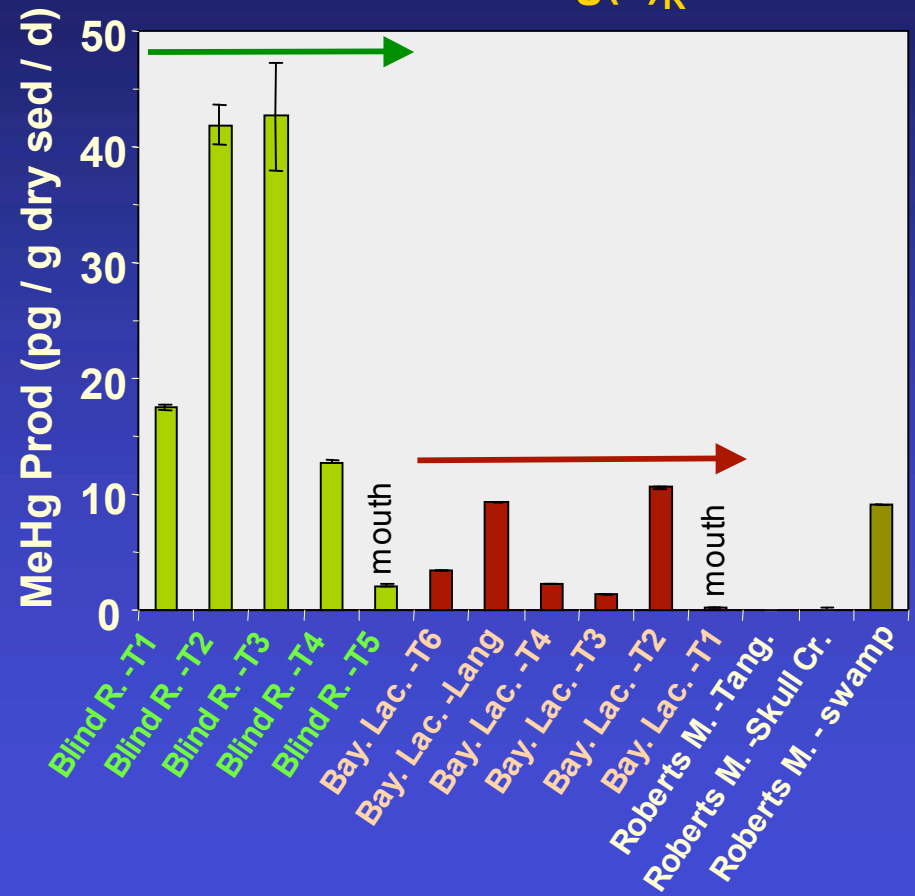
Dry Sediment Rates (surface 0-2 cm interval)

$^{203}\text{Hg(II)}$ -Methylation
Rate Constant



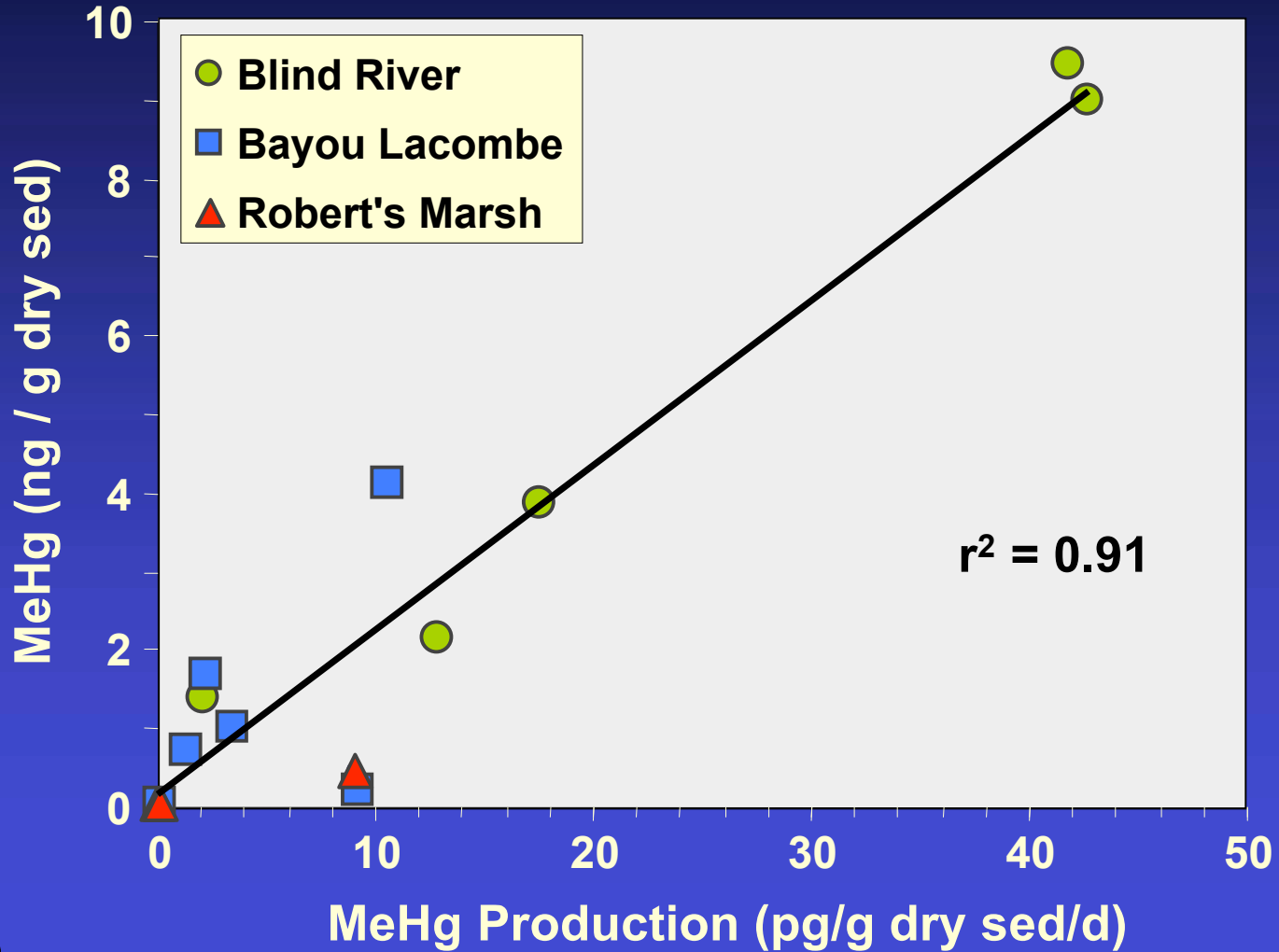
MeHg Production Rate

$$\text{MP} = K \times \text{Hg(II)}_R$$





MeHg Production based on REACTIVE Hg(II) vs MeHg concentration (dry wt.)



Key Questions

- What are the current trends wrt MeHg production (Hg(II)_{R} , S, and Fe chemistry) with increasing salinity in the ponds (is there an optimum)?
- How will these parameters change as we convert the salt ponds to tidal wetland and managed ponds?
- How does Hg currently cycle under the target wetland conditions? (i.e. What is the background?)
- How will deepening Alviso Slough impact Hg(II)_{R} concentrations in the ponds and in South SFB? And for how long?

Take Home Message

- South SFB & Salt Ponds have elevated Hg in sediments, water and biota (top predators)
- Limited understanding of the specific underlying processes involved in Hg cycling in this Bay region
- MeHg Production is a function of microbial activity and availability of reactive Hg(II)
- MeHg production / degradation rates vary widely (but predictably) among and within ecosystem sub-habitats
- Plants can have multiple and substantial effects on Hg cycling.
- Reactive-Hg(II) varies with sediment chemistry (organics, redox, ...) and is a small fraction of Hg_T ($\leq 5\%$)
- Wetland Restoration: Success will depend on understanding the controls on Hg cycling in various sub-habitats that comprise the project area